

**8.2****THE NATURE OF COVALENT BONDING****Section Review****Objectives**

- State a rule that usually tells how many electrons are shared to form a covalent bond
- Describe how electron dot formulas are used
- Predict when two atoms are likely to be joined by a double or a triple covalent bond
- Distinguish between a single covalent bond and other covalent bonds
- Describe how the strength of a covalent bond is related to its bond dissociation energy
- Describe how resonance structures explain bonding
- Identify some exceptions to the octet rule

**Vocabulary**

- |                        |                            |                            |
|------------------------|----------------------------|----------------------------|
| • single covalent bond | • double covalent bonds    | • polyatomic ion           |
| • structural formulas  | • triple covalent bonds    | • bond dissociation energy |
| • unshared pairs       | • coordinate covalent bond | • resonance structures     |

**Part A Completion**

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

- When atoms share electrons to gain the 1 configuration of a noble gas, the bonds formed are 2. A 3 pair of valence electrons constitutes a 4 covalent bond. Pairs of valence electrons that are not shared between atoms are called 5. Sometimes two or three pairs of electrons may be shared to give 6 covalent bonds. In some cases, only one of the atoms in a bond provides the pair of bonding electrons; this is a 7. 8 is required to break covalent bonds between atoms. The total energy required to break the bond between two covalently bonded atoms is known as the 9.
- When it is possible to write two or more valid electron dot formulas for a molecule or ion, each formula is referred to as a 10.
1. \_\_\_\_\_
  2. \_\_\_\_\_
  3. \_\_\_\_\_
  4. \_\_\_\_\_
  5. \_\_\_\_\_
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  9. \_\_\_\_\_
  10. \_\_\_\_\_

# The Nature Of Covalent Bonding Worksheet

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