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8.2 The Nature of Covalent Bonding

Essential Understanding | Covalent bonds form when atoms share electrons.

Lesson Summary

The Octet Rule in Covalent Bonding

Covalent compounds are most stable when each atom has eight electrons.

- Single, double, and triple covalent bonds depend on the number of pairs of electrons shared between two atoms.
- Atoms form double or triple covalent bonds if they can attain a noble gas structure by doing so.

Type of Covalent Bond	Attributes	
Single	One shared electron pair with one electron from each atom	
Double	Two shared electron pairs with two electrons from each atom	
Triple	Three shared electron pairs with three electrons from each atom	

Coordinate Covalent Bonds In a coordinate covalent bond, one atom contributes both electrons in the bonding pair.

- One atom may contribute a pair of unshared electrons to a bond to give both atoms an inert gas configuration.
- Coordinate covalent bonds can also occur in polyatomic ions, such as NH₄*.

Exceptions to the Octet Rule Some molecules have fewer, or more, than a complete octet of valence electrons.

- Molecules that have an odd number of total valence electrons cannot satisfy the octet rule.
- Some molecules that have an even number of valence electrons may also fail to follow the octet rule.

The Nature Of Covalent Bonding 16 Worksheet

Anil N. Netravali, K. L. Mittal

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Fundamentals of Biomaterials Vasif Hasirci, Nesrin Hasirci, 2018-11-26 This text for advanced undergraduate and graduate students covers the fundamental relationships between the structure and properties of materials and biological tissues The successful integration of material and biological properties shape and architecture to engineer a wide range of optimized designs for specific functions is the ultimate aim of a biomaterials scientist Relevant examples illustrate the intrinsic and tailored properties of metal ceramic polymeric carbon derived composite and naturally derived biomaterials Fundamentals of Biomaterials is written in a single voice ensuring clarity and continuity of the text and content As a result the reader will be gradually familiarized with the field starting with materials and their properties and eventually leading to critical interactions with the host environment Classical and novel examples illuminate topics from basic material properties to tissue engineering nanobiomaterials and guided tissue regeneration This comprehensive and engaging text integrates materials and biological properties to understand biomaterials function and design provides the basics of biological tissue components and hierarchy includes recent topics from tissue engineering and guided tissue regeneration to nanoarchitecture of biomaterials and their surfaces contains perspectives case studies from widely recognized experts in the field features chapter ending summaries to help readers to identify the key take home messages Student Study Guide Peter L. Kresan, Reed Mencke, Frank Press, Raymond Siever, 2003-09-25 This reconceptualization of the text Understanding Earth reflects the fundamental changes in the field of physical geology over the past several years **Functional Tactile** Sensors Ye Zhou, Ho-Hsiu Chou, 2021-01-20 Functional Tactile Sensors Materials Devices and Integrations focuses on the subject of novel materials design and device integration of tactile sensors for functional applications. The book addresses the design materials characteristics device operation principles specialized device application and mechanisms of the latest reported tactile sensors. The emphasis of the book lies in the materials science aspects of tactile sensors understanding the relationship between material properties and device performance It will be an ideal resource for researchers working in materials science engineering and physics Includes the latest advances and recent developments in tactile sensors for artificial intelligence applications Reviews the relationship between materials properties and device performance Addresses materials and device design strategies for targeted sensing applications Cracking the SAT II Chemistry Theodore Silver, Princeton Review (Firm), 2005-03 Why The Princeton Review 1 We Know the SAT Chemistry Subject Test The experts at The Princeton Review have spent many years researching the SAT Chemistry Subject Test as well as numerous other standardized tests We re confident this guide delivers the most current and complete information you need to ace this test 2 We Get Results Our inventive approach to standardized test taking has revolutionized the test prep industry and made our courses and tutoring for the SAT and SAT Subject Tests the most popular anywhere The same proven techniques we teach in our courses are also covered in this book 3 We Understand Students Each year we help more than two million students score

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computational modelling and numerical simulation tools and confirmed by modern experimental techniques It begins by summarizing basic theoretical methods then giving both a theoretical and experimental treatment of how alkali metal clusters develop into nanostructures as influenced by the cluster's magic number of atoms. The book continues with a discussion of atomic clusters and nanostructures focusing primarily on boron and carbon exploring in detail the one two and three dimensional structures of boron and carbon and describing their myriad potential applications in nanotechnology from nanocoating and nanosensing to nanobatteries with high borophene capacity. The broad discussion of computational modelling as well as the specific applications to boron and carbon make this book an essential reference resource for materials scientists in this field of research

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