# **Answers to Topic questions**

## Topic 1 Basic agricultural chemistry Answers

1

- a Molecule: can be broken down chemically into the elements of which they are composed. Example: carbon dioxide, water, sodium chloride, etc.
- b Isotopes: atoms of the same element with different atomic weights. Example: chlorine.
- Valence electron: outermost electrons around an atom.
- 2 The atom has a small, positively charged core called the nucleus. The atom is electrically neutral and consists of: 1) electrons negatively charged particles contained in a large outer volume called an electron cloud; 2) protons positively charged particles concentrated in a small inner volume that contain most of the mass of the atom; 3) neutrons electrically neutral and occur in the nucleus.

3

- a alkali-earth metal
- b fluorine / chlorine / bromine / iodine

4

- a covalent bonding
- b oxidation number
- c chemical formula
- d polar
- e hydrogen bonding

5 b

6 Hydrogen bonds that hold water molecules together are quite strong. But covalent bonds between atoms in a water molecule are stronger. Thus hydrogen bonds often break and reform which provide water's ability to flow.

7

- a carbon, oxygen
- b 2 of Fe, Mn, B, Mo, Cu, Zn, Cl, Co
- 8 Plants use CO<sub>2</sub> and H<sub>2</sub>O to build complex organic compounds. They are excreted by animals during respiration.

9

- a alcohol
- b aldehyde
- c amide
- d sugar

# **Topic Test Answers E202environmental Science**

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#### **Topic Test Answers E202environmental Science:**

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