Answer each item for 2 points each.

Define the following terms.

- 1. What is an increase?
- 2. Define atomic mass in terms of substrastic particles.
- 3. What is the Author Principle?
- 4. What is the dual nature of electrons?
- 5. What is STP?
- 6. Which insteps is the basis for the mole concept and relative atomic manal?

Explain the following chemical concepts.

- Apply kinetic energy to explain how solids differ from liquids.
- B. In air matter? Explain:
- 9. Why do atoms form objected bonds?
- 10. When Li and O bond, is it ionic, covalent, or metallic? Explain.
- 11. What is the relativeship between wavelength and frequency?
- (2). What is the normal pHI range for approva solutions?
- Disscribe the equivalence point of an acid/hour tileation.
- 14. Provide an example of a dipostic sold.
- 15. What is the difference between a suithle and sulfate ion?
- 56. What is the JORC of pure water at 25 C?
- 17. In a solution of the hydronium ion concentration is less than the hydroxide ion concentration, is the solution soldie, fessio, or neutral?

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Anthony S. Fauci

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