

## 7-1 Review and Reinforcement

### Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative  
positive  
electron

noble gas  
atoms  
ion

ion  
ion  
cation

electron configuration  
polyatomic  
empirical formula

1. An ionic bond is an attraction between oppositely charged \_\_\_\_ **ions** \_\_\_\_.
2. Anionic bond is \_\_\_\_ **negative** \_\_\_\_ change.
3. An ionic bond is formed by losing or gaining \_\_\_\_ **electrons** \_\_\_\_.
4. The \_\_\_\_ **metal** \_\_\_\_ cation cation that cation tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form a bond bond, both have negative the electron configuration of \_\_\_\_ **noble gas** \_\_\_\_.
6. The atoms that make up \_\_\_\_ **polyatomic** \_\_\_\_ ions are bonded together by covalent bonds.
7. The \_\_\_\_ **crystalline** \_\_\_\_ method can be used to write the formula for an ionic compound.
8. The \_\_\_\_ **empirical formula** \_\_\_\_ of a compound denotes the ratio of ions in the compound. If the compound is ionic, write "ion" if it is ionic. Change the subscript when it needs to make it true. Write your answer on the line.

\_\_\_\_ **ionic** \_\_\_\_

\_\_\_\_ **ion** \_\_\_\_

\_\_\_\_ **ion** \_\_\_\_

\_\_\_\_ **polyatomic** \_\_\_\_

\_\_\_\_ **crystal** \_\_\_\_

\_\_\_\_ **ion (polyatomic)** \_\_\_\_

9. Calcium bromide is composed of calcium ion (gaining two electrons).

10. A cation has a positive charge.

11. A binary ionic compound contains only one kind of cation and one kind of anion.

12. Substances have number of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.

14. A great deal of energy is produced when an ionic compound is formed.

# Stars Section 1 Reinforcement Answers

**David Kirk**



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