

1. The table shows some information about solids, liquids and gases.

Complete the table.

| state  | distance between particles | movement of particles          | forces between particles | shape              |
|--------|----------------------------|--------------------------------|--------------------------|--------------------|
| solid  | close together             | .....<br>.....                 | .....<br>.....           | fixed shape        |
| liquid | close together             | move slowly in all directions  | .....<br>.....           | shape of container |
| gas    | .....<br>.....             | move quickly in all directions | very weak                | .....<br>.....     |

2. This question is about liquids.

(a) When a liquid is put into a beaker it takes on the shape of the beaker.

Why do liquids take on the shape of the beaker?

Tick (✓) the boxes next to **two** correct answers.

Liquids have no fixed shape.

☐

Liquids have no fixed volume.

☐

Liquid particles only vibrate.

☐

The particles in a liquid expand.

☐

The particles in a liquid move around.

☐

(b) Pierre investigates the boiling point of water.

- He heats up some pure water in a beaker until it boils.
- Pierre measures the boiling point of the pure water.
- He does this two more times using new samples of pure water.
- Pierre then does this three more times using salt water.

The table shows some of his results.

| type of water | boiling point in °C |     |     |
|---------------|---------------------|-----|-----|
| pure water    | 100                 | 100 | 100 |
| salt water    | 102                 | 103 |     |

(i) There is one result missing. The thermometer shows Pierre's missing result.

# States Of Matter Section 13 1 Gases Worksheet Answers

**SB Merriam**



## **States Of Matter Section 13 1 Gases Worksheet Answers:**

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