# Simple Stateminmetry Lap

Name	Period	Date
CONTRACTOR OF THE PROPERTY OF		The Control of the Co

To conclude the theoretical yield in a reaction, perform the reaction, and conclude the percent yield.

Have provided the class, we've been talking about stricksometry, the prediction of reaction masses using chemical equations. In this lab, you will be mixing mosts acid with codings hydroxide to form codings accrate and water. The equations for this reaction is:

You will start with a known quentity of scotic acid and codemn by drawide and will gredict the mass of sodium scottes formed. You will then measure the smooth of sodium scottes actually graduced by boiling the reaction mixture to drawns. By calculating your percent staid, you can determine your accuracy!

#### Property Chamberry

Is boiling the writer to scalate the sodium acetate the best way to measure the amount of sodium acetate formed?

Explain why or why and

#### (December)

Manager angery 100 ml. manted	1
Manual special series	
Министипации причини	
Moscar 200 ml. season was enable seems.	

#### Remove Characters

To provide these produces and YOUF date, has the party or management that over a suggested in the task?

1. Under the equation below, write the measured mass that you used (of each reaction in greats) during the lab.

- Using your measured man of the scenic sold, colonies the theoretical yield of sodium scenes. (Show your work. factor label medical?)
- 3. Calculate the screet yield of codings according the volum in the data section.
- 4. Calculate the percent yield of notions scenes. Show your world
- According to the Law of Conservation of Moss, we cannot crown stone. Therefore, it is suppossible to achieve more
  than 100% yield. What are an experimental errors could have accounted to come some lab groups to have more than
  100% yield? Think? Shanna error is 2007 on acceptable survey?
- It is highly making that you would ever achieve 100% yould. Moor often, into groups full short of 100%. What was experimental errors could have occurred to decrease yield? Think! Not human error?

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