

1. What explains the very high melting and boiling point of water

- |                         |  |                         |   |
|-------------------------|--|-------------------------|---|
| <input type="radio"/> A | Strong dipole-dipole bonds between water molecules   | <input type="radio"/> B | Strong hydrogen bonds between water molecules |
| <input type="radio"/> C | Dispersion forces which are present in all molecules | <input type="radio"/> D | Asymmetrical shape of the polar bonds.        |

2. Does CH<sub>4</sub> have hydrogen bonding?

- |                         |    |                         |     |
|-------------------------|----|-------------------------|-----|
| <input type="radio"/> A | no | <input type="radio"/> B | yes |
|-------------------------|----|-------------------------|-----|

3.  At higher temperatures

- |                         |   |                         |                                    |
|-------------------------|---|-------------------------|------------------------------------|
| <input type="radio"/> A | a gas contracts                         | <input type="radio"/> B | particles in an object move faster |
| <input type="radio"/> C | particles in an object have less energy |                         |                                    |

4.  A gas

- |                         |   |                         |                           |
|-------------------------|---|-------------------------|---------------------------|
| <input type="radio"/> A | has a definite shape but no definite volume | <input type="radio"/> B | has fast-moving particles |
| <input type="radio"/> C | has a definite volume but no definite shape |                         |                           |

5. Define a liquid.

- |                         |                                   |                         |                |
|-------------------------|-----------------------------------|-------------------------|----------------|
| <input type="radio"/> A | Definite volume, indefinite shape | <input type="radio"/> B | Found in stars |
|-------------------------|-----------------------------------|-------------------------|----------------|

# States Of Matter And Intermolecular Forces Quiz

**Arshad Iqbal**



## States Of Matter And Intermolecular Forces Quiz:

Chemistry Class 11 Dr. S C Rastogi, ,Er. Meera Goyal,2022-09-29 Syllabus Unit I Some Basic Concepts of Chemistry Unit II Structure of Atom Unit III Classification of Elements and Periodicity in Properties Unit IV Chemical Bonding and Molecular Structure Unit V States of Matter Gases and Liquids Unit VI Chemical Thermodynamics Unit VII Equilibrium Unit VIII Redox Reactions Unit IX Hydrogen Unit X s Block Elements Alkali and Alkaline earth metals Group 1 and Group 2 Elements Unit XI Some p Block Elements General Introduction to p Block Elements Unit XII Organic Chemistry Some Basic Principles and Techniques Unit XIII Hydrocarbons Classification of Hydrocarbons Unit XI V Environmental Chemistry Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table      Class 11-12 Chemistry MCQ (Multiple Choice Questions) Arshad Iqbal,2019-05-17 The Class 11 12 Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF College Chemistry MCQ PDF Download Quiz Questions Chapter 1 6 Practice Tests with Answer Key 11th 12th Grade Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Class 11 12 Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry MCQ PDF book helps to practice test questions from exam prep notes The Class 11 12 Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Class 11 12 Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Class 11 12 Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book Grade 11 12 Chemistry MCQs Chapter 1 6 PDF includes college question papers to review practice tests for exams Class 11 12 Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Mock Tests Chapter 1 6 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure MCQ Chapter 2 Basic Chemistry MCQ Chapter 3 Chemical Bonding MCQ Chapter 4 Experimental Techniques MCQ Chapter 5 Gases MCQ Chapter 6 Liquids and Solids MCQ The Atomic Structure MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model defects charge to mass ratio of electron discovery of electron discovery of neutron discovery of proton dual nature of matter electron charge electron distribution electron radius and energy derivation electron velocity electronic

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