

1. What explains the very high melting and boiling point of water

- | | |
|--|--|
| <input type="radio"/> A Strong dipole-dipole bonds between water molecules | <input type="radio"/> B Dispersion forces which are present in all molecules |
| <input type="radio"/> C Strong hydrogen bonds between water molecules | <input type="radio"/> D Asymmetrical shape of the polar bonds. |

2. Does HCl have hydrogen bonding?

- | | |
|----------------------------|-----------------------------|
| <input type="radio"/> A no | <input type="radio"/> B yes |
|----------------------------|-----------------------------|

3. Does CH₄ have hydrogen bonding?

- | | |
|-----------------------------|----------------------------|
| <input type="radio"/> A yes | <input type="radio"/> B no |
|-----------------------------|----------------------------|

4. All molecules have London forces between them, but dipole-dipole and hydrogen bonding are so much stronger that when they are present we can ignore London forces. Which of these has ONLY London forces?

- | | |
|---|--|
| <input type="radio"/> A I ₂ | <input type="radio"/> B OCl ₂ |
| <input type="radio"/> C NH ₃ | <input type="radio"/> D SH ₂ |

5. Which substance would have the weakest intermolecular forces of attraction?

- | | |
|---|--|
| <input type="radio"/> A CH ₄ | <input type="radio"/> B H ₂ O |
| <input type="radio"/> C NaCl | <input type="radio"/> D MgF ₂ |

6. Intermolecular forces for: NH₃



- | | |
|---------------------------------------|--|
| <input type="radio"/> A Dipole dipole | <input type="radio"/> B Dispersion Force |
|---------------------------------------|--|

States Of Matter And Intermolecular Forces Test

JE Gale



States Of Matter And Intermolecular Forces Test:

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States Of Matter A. Goel,2006 This book has been written for the students of under graduate and post graduate level of the various universities in India A special feature of the book is that the text has been illustrated with a large number of line diagrams and the data presented in the form of numerous tables for reference and comparison In the preparation of text standard works and review by renowned author have been freely consulted and the reference given chapter wise At the end of the book will be found useful by those who wish to make a more detailed study of the topics discussed Contents Liquid State Solids State Gaseous State

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(Chemistry) Physical Chemistry: States of Matter and Ionic Equilibrium Dr. MD. Faiyaz Ahsan,Dr. Nilima,2024-01-01 Buy Latest Chemistry Physical Chemistry States of Matter and Ionic Equilibrium e Book in English Edition for B Sc 2nd Semester Bihar State By Thakur publication

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Sevian, 2013-07-09 Bringing together a wide collection of ideas reviews analyses and new research on particulate and structural concepts of matter Concepts of Matter in Science Education informs practice from pre school through graduate school learning and teaching and aims to inspire progress in science education The expert contributors offer a range of reviews and critical analyses of related literature and in depth analysis of specific issues as well as new research Among the themes covered are learning progressions for teaching a particle model of matter the mental models of both students and

teachers of the particulate nature of matter educational technology chemical reactions and chemical phenomena chemical structure and bonding quantum chemistry and the history and philosophy of science relating to the particulate nature of matter The book will benefit a wide audience including classroom practitioners and student teachers at every educational level teacher educators and researchers in science education If gaining the precise meaning in particulate terms of what is solid what is liquid and that air is a gas were that simple we would not be confronted with another book which while suggesting new approaches to teaching these topics confirms they are still very difficult for students to learn Peter Fensham Emeritus Professor Monash University Adjunct Professor QUT from the foreword to this book

S. Chand's Principles Of Physics For XI V. K Mehta & Rohit Mehta, The Present book S Chand s Principle of Physics is written primarily for the students preparing for CBSE Examination as per new Syllabus Simple language and systematic development of the subject matter Emphasis on concepts and clear mathematical derivations

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The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts.

The Secret History is a brilliant and suspenseful novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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to the following drawing using the same general tolerances as those specified on the rocker arm

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