



**11** How does the arrangement of particles in a solid change when it melts? What happens to the energy of the particles when it melts?

**10** How does the arrangement of particles in a solid change when it melts? What happens to the energy of the particles when it melts?

**9** What is the change of state when a solid melts?

**8** What is the change of state when a solid melts?

**7** How does the arrangement of particles in a solid change when it melts?

## PARTICLE THEORY AND STATES OF MATTER

Particle theory explains the properties of matter in terms of the arrangement of particles. The particles in matter are constantly moving and colliding. The arrangement of particles in a solid is fixed, but in a liquid it is more random. In a gas, the particles are far apart and move freely.

**States of Matter:** Matter can exist in three primary states: solid, liquid, and gas. Solids have a fixed shape and volume. Liquids have a fixed volume but take the shape of their container. Gases have neither a fixed shape nor volume.

**Changes of State:** Changes of state occur when the arrangement of particles changes. Melting is the change from solid to liquid, and boiling is the change from liquid to gas. These changes involve the addition or removal of energy.

**Particle Arrangement:** In a solid, particles are packed closely together. In a liquid, they are more spread out. In a gas, they are very far apart and move rapidly.



**1** What is particle theory?

**2** What are the three states of matter?

**3** How does the arrangement of particles in a solid change when it melts?

**4** How does the arrangement of particles in a liquid change when it boils?

**5** What happens when a solid melts?

**6** How does the arrangement of particles in a gas change when it condenses?

Particle Theory and States of Matter  
Guided Reading

# States Of Matter Guided Reading Answer Key

**LP Steffe**



## **States Of Matter Guided Reading Answer Key:**

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