

PROPERTIES OF WATER

Reading Comprehension

Grades 8-12

NAME: _____ **PER:** _____ **DATE:** _____

PROPERTIES OF WATER

High Capacity and High Heat of Vaporization

(21) Heat capacity refers to the ability of a substance to absorb heat. Water has a high capacity to absorb heat, which means it can absorb a lot of heat without its temperature rising very quickly. This means that water can absorb a lot of heat from the sun or from a fire, and it can release that heat back out into the environment. This property is what allows water to regulate Earth's climate. Water's high heat of vaporization means that it takes a lot of energy to turn water into a gas. This means that water can absorb a lot of heat from the sun or from a fire, and it can release that heat back out into the environment. This property is what allows water to regulate Earth's climate. Water's high heat of vaporization means that it takes a lot of energy to turn water into a gas. This means that water can absorb a lot of heat from the sun or from a fire, and it can release that heat back out into the environment. This property is what allows water to regulate Earth's climate.

Adhesion and Cohesion

(22) Hydrogen bonds between water molecules give water the properties of adhesion and cohesion. Adhesion is the attraction between water molecules and other substances. Cohesion is the attraction between water molecules and other water molecules. This property is what allows water to move up a plant stem. Water's ability to move up a plant stem is due to its high heat of vaporization. Water's high heat of vaporization means that it takes a lot of energy to turn water into a gas. This means that water can absorb a lot of heat from the sun or from a fire, and it can release that heat back out into the environment. This property is what allows water to regulate Earth's climate.

Capillary Action in Plants

(23) Capillary action occurs due to the combined effects of adhesion and cohesion. Capillary action is the ability of water to move up a plant stem. Water's ability to move up a plant stem is due to its high heat of vaporization. Water's high heat of vaporization means that it takes a lot of energy to turn water into a gas. This means that water can absorb a lot of heat from the sun or from a fire, and it can release that heat back out into the environment. This property is what allows water to regulate Earth's climate.

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PROPERTIES OF WATER

Hydrogen Bonds

(24) Water is a polar molecule. It is made up of two hydrogen atoms and one oxygen atom. The oxygen atom is more electronegative than the hydrogen atoms, which means it attracts electrons more strongly. This creates a partial negative charge on the oxygen atom and a partial positive charge on the hydrogen atoms. This property is what allows water to form hydrogen bonds. Hydrogen bonds are weak attractions between water molecules. They are responsible for many of water's unique properties, such as its high heat of vaporization and its ability to move up a plant stem.

Surface Tension

(25) Because of water's cohesive nature, each molecule of water has two hydrogen bonds with other water molecules. This strong attraction between water molecules creates a force called surface tension. Surface tension is the force that holds the molecules of water together at the surface. It is responsible for many of water's unique properties, such as its ability to move up a plant stem.

Why is water a polar molecule?

(26) Water is a polar molecule because it is made up of two hydrogen atoms and one oxygen atom. The oxygen atom is more electronegative than the hydrogen atoms, which means it attracts electrons more strongly. This creates a partial negative charge on the oxygen atom and a partial positive charge on the hydrogen atoms. This property is what allows water to form hydrogen bonds.

Why is water referred to as a universal solvent?

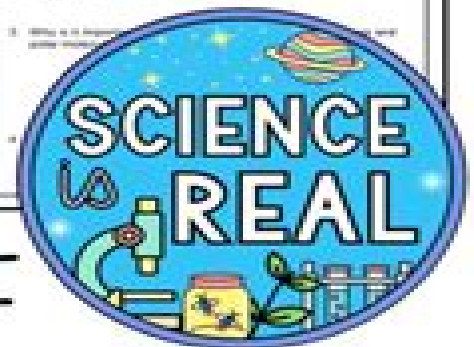
(27) Water is referred to as a universal solvent because it is able to dissolve many different substances. This is due to its polar nature. Water's polarity allows it to form hydrogen bonds with many different substances, which allows it to dissolve them.

Why is it important to understand the properties of water?

(28) It is important to understand the properties of water because they are responsible for many of the unique properties of water. These properties are what allow water to move up a plant stem, form hydrogen bonds, and create surface tension. Understanding these properties is essential for understanding many of the processes that occur in nature.

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