PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

C 2.2 ______1. cation a. a positively charged ion C2.2 8 2. electronegativity b. an element whose highest occupied s or p sublevels are partially filled C.2.5 3. inner transition metal c. the energy required to remove an electron from an atom C.2.6 4. alkali earth metal d. good conductor of heat and electric current. C.2.6 h S. noble gas e. any element whose highest occupied s sublevel and a nearby d sublevel contain electrons C.2.5 6. representative element f. any nonmetal in group 7A c 2 2 ______ 7. metal g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) C.2.6 f 8. halogen h. an element in which the highest occupied s and p sublevels are completely filled C2.5 ______9. transition metal i. an element in group 2A C.2.2 c 10. ionization energy an element whose highest s sublevel and a nearby f sublevel contain electrons

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- C.2.6 __b __11. The alkali metals do not include
 a. l.i c. Na
 b. Ca d. Rb
- C.2.6 _____12. The element iodine, I, is a a. period S alkali metal b. period 4 halogen

c. period 5 halogen d. period 5 transition metal

World Of Chemistry Chapter Test Answers

RS Peters

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The World of Chemistry Jones, Jones Staff, 1996 World of Chemistry Melvin D. Joesten, James L. Wood, 1999 The quality of human life is described in terms understanding to the non science major in this application oriented chemistry textbook This paperback offers a more concise less theoretical less mathematical version of the hardback version WORLD OF **CHEMISTRY Second Edition** Student Topic Outline and Study Guide to Accompany World of Chemistry, Essentials, Joesten/Netterville/Wood James O. Schreck, 1993 Concepts of Earth Science & Chemistry Parent Lesson Plan John Hudson Tiner, 2013-08-26 Concepts of Earth and Chemistry Course Description This is the suggested course sequence that allows one core area of science to be studied per semester You can change the sequence of the semesters per the needs or interests of your student materials for each semester are independent of one another to allow flexibility Semester 1 Earth Blending a creationism perspective of history with definitions of terms and identification of famous explorers scientists etc this book gives students an excellent initial knowledge of people and places encouraging them to continue their studies in depth Semester 2 Chemistry Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical Survey of Science History & Concepts Parent Lesson Plan ,2013-08-01 Survey of Science History information Concepts Course Description Students will study four areas of science Scientific Mathematics Physics Biology and Chemistry Students will gain an appreciation for how each subject has affected our lives and for the people God revealed wisdom to as they sought to understand Creation Each content area is thoroughly explored giving students a good foundation in each discipline Semester 1 Math and Physics Numbers surround us Just try to make it through a day without using any It s impossible telephone numbers calendars volume settings shoe sizes speed limits weights street numbers microwave timers TV channels and the list goes on and on The many advancements and branches of mathematics were developed through the centuries as people encountered problems and relied upon math to solve them It's amazing how ten simple digits can be used in an endless number of ways to benefit man The development of these ten digits and their many uses is the fascinating story in Exploring the World of Mathematics Physics is a branch of science that many people consider to be too complicated to understand John Hudson Tiner puts this myth to rest as he explains the fascinating world of physics in a way that students can comprehend Did you know that a feather and a lump of lead will fall at the same rate in a vacuum Learn about the history

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provides an introduction to organic chemistry for students who require the fundamentals of organic chemistry as a requirement for their major It is most suited for a one semester organic chemistry course In an attempt to highlight the relevance of the material to students the authors place a strong emphasis on showing the interrelationship between organic chemistry and other areas of science particularly the biological and health sciences. The text illustrates the use of organic chemistry as a tool in these sciences it also stresses the organic compounds both natural and synthetic that surround us in everyday life in pharmaceuticals plastics fibers agrochemicals surface coatings toiletry preparations and cosmetics food Normal Instructor and additives adhesives and elastomers Mathematics & Science in the Real World ,2000 **Teachers World** .1923 English Mechanic and World of Science, 1889 Scientific American Biology for a Changing World Michele Shuster, Janet Vigna, Gunjan Sinha, Matthew Tontonoz, 2011-02-25 To view sample chapters and more information visit www whfreeman com SABiologyPreview All of us involved in science education understand the importance of scientific literacy How do we get the attention of a nonscientist And if we can get it how do we keep it not only for the duration of the course or the chapter in a textbook but beyond How do we convey in our courses and our textbooks not just what we know but also how science is done These are the challenges we hope to address with our new series of textbooks specifically for the nonscientist With this series W H Freeman and Scientific American join forces not just to engage nonscientists but to equip them critical life tools

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