

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

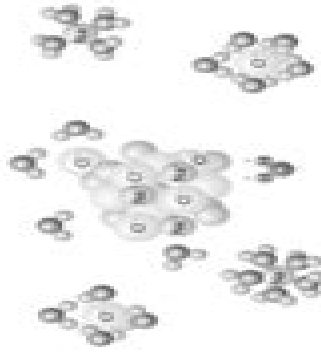
Use each of the terms below just once to complete the passage.

immiscible	liquid	solution	solute	solution
insoluble	miscible	solute	solvent	

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Study Guide For Content Master 15 Key

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Pathophysiology Kathryn L. McCance, RN, PhD, Sue E. Huether, RN, PhD, 2014-01-14 With easy to read in depth descriptions of disease disease etiology and disease processes Pathophysiology The Biologic Basis for Disease in Adults and Children 7th Edition helps you understand the most important and the most complex pathophysiology concepts More than 1 200 full color illustrations and photographs make it easier to identify normal anatomy and physiology as well as alterations of function This edition includes a NEW Epigenetics and Disease chapter along with additional What's New boxes highlighting the latest advances in pathophysiology Written by well known educators Kathryn McCance and Sue Huether and joined by a team of expert contributors this resource is the most comprehensive and authoritative pathophysiology text available Over 1 200 full color illustrations and photographs depict the clinical manifestations of disease and disease processes more than in any other pathophysiology text A fully updated glossary includes 1 000 terms and makes lookup easier by grouping together similar topics and terms Outstanding authors Kathryn McCance and Sue Huether have extensive backgrounds as researchers and instructors and utilize expert contributors consultants and reviewers in developing this edition Chapter summary reviews provide concise synopses of the main points of each chapter Consistent presentation of diseases includes pathophysiology clinical manifestations and evaluation and treatment Lifespan content includes ten separate pediatric chapters and special sections with aging and pediatrics content Algorithms and flowcharts of diseases and disorders make it easy to follow the sequential progression of disease processes Nutrition and Disease boxes explain the link between concepts of health promotion and disease EXTENSIVELY Updated content reflects advances in pathophysiology including tumor biology invasion and metastases the epidemiology of cancer diabetes mellitus insulin resistance thyroid and adrenal gland disorders female reproductive disorders including benign breast diseases and breast cancer and a separate chapter on male reproductive disorders and cancer NEW Chapter on epigenetics and disease Additional What's New boxes highlight the most current research and clinical development

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