

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

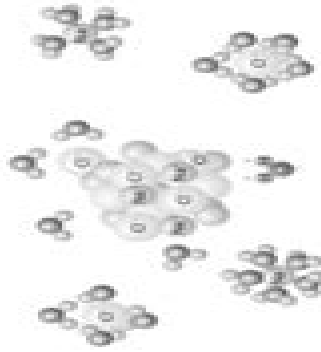
Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solvent and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Darcy West



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Study Guide & Laboratory Manual for Physical Examination & Health Assessment E-Book Carolyn Jarvis, Ann L. Eckhardt, 2023-01-02 Both a comprehensive lab manual and a practical workbook the Study Guide Laboratory Manual for Physical Examination Health Assessment 9th Edition gives you the tools you need to master physical examination and health assessment skills in the lab and in clinical practice Corresponding to the bestselling Jarvis textbook this guide features terminology reviews application activities clinical judgment questions regional write up sheets and narrative summary forms with answers to study questions at the back of the book to facilitate both learning and review The 9th edition has been thoroughly updated with a fresh focus on the Next Generation NCLEX NGN with case studies featuring new NGN question formats to prepare you not only for the skills laboratory but for success on the NCLEX and in interprofessional collaborative practice Authoritative review and guidance for laboratory experiences personally written by the textbook authors provide a seamlessly integrated study and clinical experience Consistent format includes Terminology Review Study Guide and Clinical Judgment Questions in each chapter Physical examination forms familiarize you with what you will encounter in clinical practice and offer practice in documenting the patient history and examination findings The only full color illustrated lab manual available for a nursing health assessment textbook with anatomy exercises that align with the main text NEW Clinical judgment exercises equip you for success on the Next Generation NCLEX NGN including questions with an increased focus on clinical judgment robust single episode case studies that employ the latest NGN question types and unfolding case studies which reflect the language of the NCSBN Clinical Judgment Measurement Model NEW Chapter 33 includes unfolding case studies for the NGN that present opportunities for you to practice prioritizing decision making and using clinical judgment skills NEW Increased emphasis on activities focused on higher cognitive levels Applying and above UPDATED Critical Thinking Exercises offer suggested readings based on your participation in the skills lab and discussions with your instructor UPDATED Content corresponds to the 9th edition of the Jarvis textbook and incorporates the latest research and evidence based practice [Student Study Guide With IBM® SPSS® Workbook for Research Methods for the Behavioral Sciences](#) Gregory J. Privitera, 2019-01-24 The Student Study Guide With IBM SPSS Workbook for Research Methods for the Behavioral Sciences Third Edition by Gregory J Privitera includes a review of chapter learning objectives chapter summaries and tips and cautions To help students practice their skills the guide offers quizzes and exercises accompanied by answers keys SPSS in Focus exercises with general instructions complement those in Privitera's main text Research Methods for the Behavioral Sciences Third Edition **Study Guide and Laboratory Manual for Physical Examination and Health Assessment, Canadian Edition- E-Book** Carolyn Jarvis, Ann L. Eckhardt, 2023-02-23 NEW Preface includes an overview of the NCSBN Clinical Judgment Measurement Model PN competencies and the REx PN NEW Clinical judgement exercises equip students for success on the Next Generation NCLEX NGN including questions with an increased focus on clinical judgement Updated

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The Learner-Centered Classroom Jeanette Campos, 2014-07-02
Learner centered instruction helps build dynamic classrooms in which the student takes ownership of the content and determines ways that it may be useful to him Such instruction also encourages the student to build the relationships that will allow learning to endure beyond the training event Here are a few of the questions author Jeannette Campos suggests asking yourself to effectively create a learner centered classroom As an instructor am I encouraging interaction among my learners whenever possible Is my content formatted in the most engaging way Have I created opportunities for the learner to reflect on the learning process Only in a learner centered classroom can you foster the four types of learning relationships learner to instructor learner to content learner to learner and learner to self This TD at Work will explain the four relationships within a classroom whether online or face to face the qualities of healthy high performing classroom relationships how different teaching methods for example podcasts peer review and asking questions can strengthen learner centered relationships the four part learning objectives to design learner centered instruction

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