

Chapter 12

STUDY GUIDE

Mixtures and Solutions

Section 12.1: Homogeneous and Heterogeneous Mixtures

Do what you can, when you can, as often as you can.

The characteristics below, either true or false.

- _____ **True** _____ 1. A homogeneous mixture has uniform properties and composition.
- _____ **True** _____ 2. The most common homogeneous mixture is the liquid phase solution.
- _____ **True** _____ 3. A solution is a homogeneous mixture.
- _____ **True** _____ 4. A solid solution consists of a dissolved solute in a solvent.
- _____ **True** _____ 5. Nitrogen is not a homogeneous mixture.
- _____ **False** _____ 6. An alcohol-ethanol mixture is not a solution.
- _____ **True** _____ 7. A mixture of two or more substances is a solution.
- _____ **True** _____ 8. A mixture of two or more substances is a solution.
- _____ **True** _____ 9. A mixture of two or more substances is a solution.
- _____ **True** _____ 10. A mixture of two or more substances is a solution.
- _____ **True** _____ 11. A mixture of two or more substances is a solution.
- _____ **True** _____ 12. A mixture of two or more substances is a solution.
- _____ **True** _____ 13. A mixture of two or more substances is a solution.
- _____ **True** _____ 14. A mixture of two or more substances is a solution.
- _____ **True** _____ 15. A mixture of two or more substances is a solution.
- _____ **True** _____ 16. A mixture of two or more substances is a solution.
- _____ **True** _____ 17. A mixture of two or more substances is a solution.
- _____ **True** _____ 18. A mixture of two or more substances is a solution.
- _____ **True** _____ 19. A mixture of two or more substances is a solution.
- _____ **True** _____ 20. A mixture of two or more substances is a solution.

Do what you can, when you can, as often as you can. The table below shows the characteristics of solutions, mixtures, and compounds. Place a check in the appropriate column below for each characteristic.

Characteristics of Solutions	Mixtures	Compounds	Compounds
1. A mixture of two or more substances.			
2. A mixture of two or more substances.			
3. A mixture of two or more substances.			
4. A mixture of two or more substances.			
5. A mixture of two or more substances.			
6. A mixture of two or more substances.			
7. A mixture of two or more substances.			
8. A mixture of two or more substances.			
9. A mixture of two or more substances.			
10. A mixture of two or more substances.			
11. A mixture of two or more substances.			
12. A mixture of two or more substances.			
13. A mixture of two or more substances.			
14. A mixture of two or more substances.			
15. A mixture of two or more substances.			
16. A mixture of two or more substances.			
17. A mixture of two or more substances.			
18. A mixture of two or more substances.			
19. A mixture of two or more substances.			
20. A mixture of two or more substances.			

Study Guide For Mixtures And Solutions

James E. Brady, Fred Senese



Study Guide For Mixtures And Solutions:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. *Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry* Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15
The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts. After a brief review of each section's critical ideas, students are taken through stepped-out worked examples, try-it-yourself examples, and chapter quizzes, all structured to reinforce chapter objectives and build problem-solving techniques. The solutions manual includes detailed solutions to all odd-numbered exercises in the text. Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19
This combination manual is designed to help students avoid common mistakes and understand the material better. The solutions manual section includes detailed answers and explanations to the odd-numbered exercises in the text. *Chemical Principles Study Guide/Solutions Manual* John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18
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