

# HYDROCARBONS



## Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

### a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons :

- Alkane : open chain ( $C_nH_{2n+2}$ )
- Cycloalkane : closed chain ( $C_nH_{2n}$ )

### b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond ( $C_nH_{2n}$ )
- Alkyne : triple bond ( $C_nH_{2n-2}$ )
- Alkadiene : two double bonds ( $C_nH_{2n-2}$ )



## Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of alkanes from  $C_1$ - $C_{10}$  :

Formula	Name	Formula	Name
$CH_4$	methane	$C_6H_{14}$	hexane
$C_2H_6$	ethane	$C_7H_{16}$	heptane
$C_3H_8$	propane	$C_8H_{18}$	octane
$C_4H_{10}$	butane	$C_9H_{20}$	nonane
$C_5H_{12}$	pentane	$C_{10}H_{22}$	decane

Alkene names :

$C_2H_4$  = ethylene

$C_3H_6$  = propene

# Study Guide Hydrocarbons

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## **Study Guide Hydrocarbons:**

## Reviewing **Study Guide Hydrocarbons**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Study Guide Hydrocarbons**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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