

Study Guided #3 Periodic Table Metals, Nonmetals, Metalloids

Metals	Nonmetals	Metalloids
 Metals Location: <u>Left</u> of periodic table • Shiny (luster) • Malleable/ductile (flatten without shattering/shaping) • Conducts electricity • High Boiling Point • High Melting Point • Mostly found as a solid	 Nonmetals Location: <u>Right</u> of periodic table • Dull • Brittle (shatters) • Do not conduct electricity (insulator) • Low Boiling Point • Low Melting Point • Usually found as a gas	 Metalloids Location: <u>Diagonal</u> of periodic table • Semi-conductor • Have properties of both metals and nonmetals • Usually shiny • Brittle
Trends in Atomic Radius		
Atomic Radius: Size of the atom • Across the periodic table: <u>more protons</u> • Atomic Radius: <u>decreases</u> due to adding protons, increases pull • Down the periodic table: <u>increased shells</u> • Atomic Radius: <u>increases</u> due to adding energy levels		
Ionic Radius and Naming Ions		
 Reaction: $\text{Pb} + \text{S} \rightarrow \text{Pb}^{2+} + \text{S}^{2-}$ Reaction atom: Pb^{2+} [2, 7] Reaction ion: S^{2-} [2, 8]		Ionic Radius: An atom that has gained or lost electrons, will change in atomic size • Metals tend to <u>lose</u> electrons • Valence electrons (<u>1, 2, 3, 4</u>) • Ions have <u>positive</u> charges • Ionic Radius is <u>smaller</u> than the atom • Nonmetals tend to <u>gain</u> electrons • Valence electrons (<u>4, 5, 6, 7</u>) • Ions have <u>negative</u> charge • Ionic Radius is <u>larger</u> than the atom
Trends in Ionization Energy		

Study Guide Periodic Trends

**Barron's Educational Series, Brian
Stewart**



Study Guide Periodic Trends:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement ASAP Chemistry: A Quick-Review Study Guide for the AP Exam The Princeton Review, 2019-02-12 Looking for sample exams practice questions and test taking strategies Check out our extended in depth AP chem prep guide Cracking the AP Chemistry Exam LIKE CLASS NOTES ONLY BETTER The Princeton Review s ASAP Chemistry is designed to help you zero in on just the information you need to know to successfully grapple with the AP test No questions no drills just review Advanced Placement exams require students to have a firm grasp of content you can t bluff or even logic your way to a 5 Like a set of class notes borrowed from the smartest student in your grade this book gives you exactly that No tricks or crazy stratagems no sample essays or practice sets Just the facts presented with lots of helpful visuals Inside ASAP Chemistry you ll find Essential concepts terms and functions for AP Chem all explained clearly concisely Diagrams charts and graphs for quick visual reference A three pass icon system designed to help you prioritize learning what you MUST SHOULD and COULD know in the time you have available Ask Yourself questions to help identify areas where you might need extra attention A resource that s perfect for last minute exam prep and for daily class work Topics covered in ASAP Chemistry include Atomic structure Covalent bonding intermolecular forces Thermochemistry Acids bases and more

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