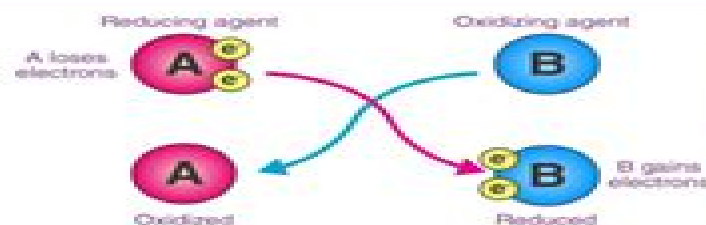


Redox reaction

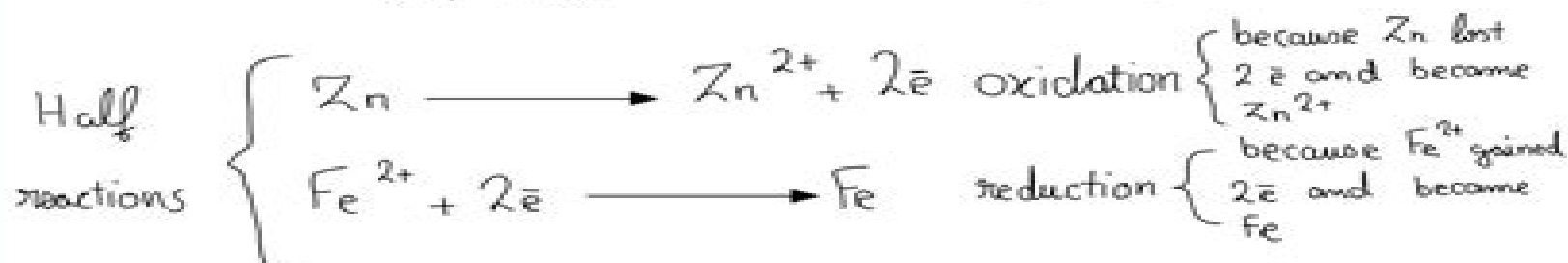
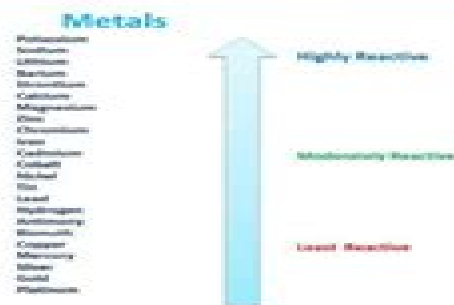
- A redox reaction is a chemical reaction where **both** oxidation and reduction take place
- An ion with positive charge is called Cations
- An ion with negative charge is called Anions
- Oxidation is a loss of electrons
- Reduction is a gain of electrons
- Not all the oxidation reactions involve oxygen

* The ion or molecule that accepts electrons is called the oxidizing agent - by accepting electrons it oxidizes other species.

* The ion or molecule that donates electrons is called the reducing agent - by giving electrons it reduced the other species.



- If a metal is more reactive than the metal corresponding to the ion there is going to be a reaction
- If a metal is less reactive than the metal corresponding to the ion there is **NOT** going to be a reaction



Study Guide For Content Mastery Redox Reactions

JS Bruner



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Magma Redox Geochemistry Roberto Moretti, Daniel R. Neuville, 2021-09-14 Explores the many facets of redox exchanges that drive magma's behavior and evolution from the origin of the Earth until today The redox state is one of the master variables behind the Earth's forming processes which at depth concern magma as the major transport agent Understanding redox exchanges in magmas is pivotal for reconstructing the history and compositional make up of our planet for exploring its mineral resources and for monitoring and forecasting volcanic activity Magma Redox Geochemistry describes the multiple facets of redox reactions in the magmatic realm and presents experimental results theoretical approaches and unconventional and novel techniques Volume highlights include Redox state and oxygen fugacity so close so far Redox processes from Earth's accretion to global geodynamics Redox evolution from the magma source to volcanic emissions Redox characterization of elements and their isotopes The American Geophysical Union promotes discovery in Earth and space science for the benefit of humanity Its publications disseminate scientific knowledge and provide resources for researchers students and professionals

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