

Introduction to Chemistry

Section 1.1 The Stories of Two Chemicals

In your textbook, read about the ozone layer.

Use each of the boxes below just once to complete the passage.

atmosphere	oxygen gas	ozone	ionized state
ionosphere	hydroxyls	ultraviolet radiation	

Earth's **OZ** _____ is made up of several layers. The air ionosphere makes up the lowest level. This layer is called the **OZ** _____. The next layer up is called the **OZ** _____. This layer contains a percentage _____.

88 _____ layer.

Chemical ozone holes **OZ** _____ is caused by ultraviolet radiation in the upper part of the atmosphere. The ozone hole is a layer around Earth, without ozone.

89 _____. Without ozone, you get more likely to get a rash or a possibly skin cancer. The lowering of the ozone layer called the **OZ** _____ is measured because without ozone all organisms on Earth are exposed to ozone from the sun's radiation.

In your textbook, read about chlorofluorocarbons.

Use each statement below, with one or more _____.

- _____ **90**. CFCs are used in the air conditioning.
- _____ **91**. CFCs are made up of carbon, fluorine, and chlorine.
- _____ **92**. All CFCs are synthetic chemicals.
- _____ **93**. CFCs usually react slowly with other chemicals.
- _____ **94**. CFCs were developed as replacements for toxic refrigerants.

Study Guide For Content Mastery Chapter 3

Clifford Lane



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