

Final Activity Series

(After 2 weeks from first activity postcard)

Notes:

After 2 weeks, students completed activity 1 and they demonstrated a chemical reaction. They reported that silver ions reacted with copper. I asked a student what happened to the silver ions. He said they were reduced. Which means they were reduced to silver metal. Which means they were reduced to silver metal. Which means they were reduced to silver metal. Which means they were reduced to silver metal.

Model 1 – Reaction of Metals



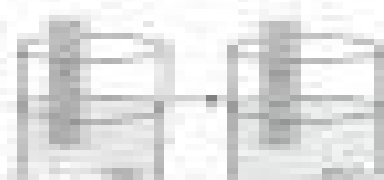
1a $Ag^+ + Cu \rightarrow Ag + Cu^{2+}$



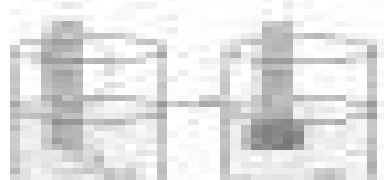
1b $Ag^+ + Cu \rightarrow Ag + Cu^{2+}$



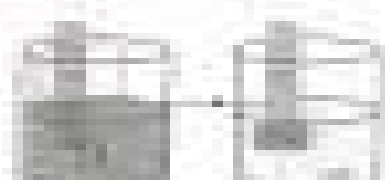
1c $Ag^+ + Cu \rightarrow Ag + Cu^{2+}$



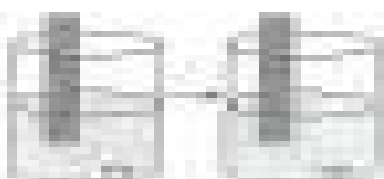
2a



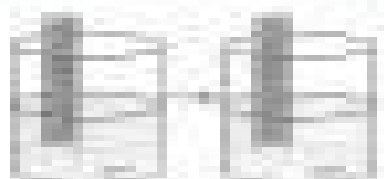
2b



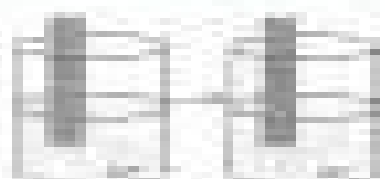
2c



3a



3b



3c

After 2 weeks, students reported evidence for a chemical reaction. They reported that silver ions reacted with copper.

The student wrote:

The Activity Series Answer Key

Disha Experts

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the 'Disha Experts' text.

The Activity Series Answer Key:

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