

## States of Matter

### Review and Reinforce

1. The bonds of attractive forces, the solid and in a liquid, and the vapors in a gas.
2. The solid would lose the water vapor because the particles in a solid are bound tightly in place. The shape of the solid will remain spread out and in a liquid because the particles in a liquid have spread out little. The shape of the vapors would continue to change and spread outwards because the particles in a gas move apart in many different directions.
3. Since the bonds of water in solid, the volume would stay the same. The volume of the liquid will increase the same. The volume of the vapors would expand to fill the size of the larger container because the particles would spread apart.
4. Matter with a definite volume and a definite shape.
5. Matter with a definite volume but no shape will be very.
6. Matter with a definite shape or volume.
7. The molecules of a liquid in flowing.
8. It refers to where the particles are not arranged in a regular pattern.
9. It refers to where the particles are arranged in a regular repeating pattern.
10. A substance that flows.
11. The bonds of an ionic solid among the molecules of a liquid that bring surface molecules closer together.

## Changes of State

### Review and Reinforce

1. from liquid
2. from liquid
3. from liquid
4. from liquid
5. from liquid
6. boiling point
7. condensation
8. evaporation
9. boiling
10. evaporation
11. melting
12. freezing
13. melting point
14. sublimation

# Skills Reinforcement Activity 20 Answers

**ReguReady**



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