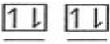
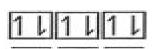
Short Answer(Show all work for calculations)

- 27. What is the orbital box diagram for Selenium?
- 28. Give the electron dot diagram for the element Iodine.
- 29. What is the electron configuration for a bismuth atom?
- 30. Give the electron configuration for a neutral atom of beryllium.
- 31. Give the electron configuration for an atom of selenium.
- 32. EXTRA CREDIT: Using the nobles gases, give the electron configuration for the following elements:
 - A) Molybdenum
 - B) Thorium
- 33. Below is the incorrect orbital diagram for Phosphorus, draw the correct orbital diagram.

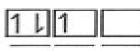






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J Spring

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Fundamentals of Ecotoxicology Michael C. Newman, 2014-12-16 An integrated analysis exploring current and relevant concepts Fundamentals of Ecotoxicology The Science of Pollution Fourth Edition extends the dialogue further from the previous editions and beyond conventional ecosystems It explores landscape regional and biospheric topics communicating core concepts with subjects ranging from molecular t *Physics, Volume 2* David Halliday, Robert Resnick, Kenneth S. Krane, 2010-04-20 Written for the full year or three term Calculus based University Physics course for science and engineering majors the publication of the first edition of Physics in 1960 launched the modern era of Physics textbooks It was a new paradigm at the time and continues to be the dominant model for all texts Physics is the most realistic option for schools looking to teach a more demanding course The entirety of Volume 2 of the 5th edition has been edited to clarify conceptual development in light of recent findings of physics education research End of chapter problem sets are thoroughly over hauled new problems are added outdated references are deleted and new short answer conceptual questions are added

MCAT General Chemistry Review Alexander Stone Macnow, 2016-07-05 The most efficient learning for the MCAT results you want Kaplan's MCAT General Chemistry Review has all the information and strategies you need to score higher on the MCAT This book features more practice than any other guide plus targeted subject review questions opportunities for self analysis a complete online center and thorough instruction on all of the general chemistry concepts necessary for MCAT success from the creators of the 1 MCAT prep course page 4 of cover Chemical Structure and Reactivity James Keeler, Peter Wothers, 2013-11 Why do certain substances react together in the way that they do What determines the shape of molecules And how can we predict whether a particular reaction will happen at all Such guestions lie at the heart of chemistry the science of understanding the composition of substances their reactions and properties Though introductory chemistry is often broken into three sections inorganic organic and physical the only way for students to fully understand the subject is to see it as a single unified whole Chemical Structure and Reactivity rises to the challenge of depicting the reality of chemistry Offering a fresh approach to the subject by depicting it as a seamless discipline the text shows how organic inorganic and physical concepts can be blended together in order to achieve the common goal of understanding chemical systems With a lively and engaging writing style enhanced by vivid illustrations only Chemical Structure and Reactivity makes teaching chemistry with an integrated approach possible Special Features The only introductory text to take a truly integrated approach in explaining the fundamentals of chemistry Fosters an orbital based understanding of reactions with clear curly arrow mechanistic detail throughout A two part structure allows flexibility of use Part I lays down the core of the subject while Part II describes a series of relatively standalone topics which can be selected to fit a particular course Numerous concepts are illustrated with fully cross referenced custom developed online modules enabling students to develop an understanding through active learning Self test exercises embedded in the text with solutions at the end of each chapter and extensive question sets encourage hands on learning to help students master the subject and gain confidence The Online Resource Centre features a range of additional resources for both students and registered adopters of the book New to this Edition A new chapter on symmetry has been added to Part I Discussions of organometallic chemistry spectroscopy and molecular geometry have been expanded Cross references from Part I to Part II have been increased to make the links between core concepts and more advanced topics clearer More self test questions and exercises have been provided

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