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molecule - answer√√two nonmetal atoms combined form a covalent compound called a

diatomic molecules - answer√√H2.O2.F2

Matter - answer√√ Anything that has mass and takes up space

Element - answer√√A substance that cannot be broken down into simpler substances

Compound - answer√√A pure substance made of two or more elements CHEMICALLY combined.

Compound - answer ✓ ✓ Can be decomposed by chemically

Mixture - answer√√A PHYSICAL blend of two or more components

(aq) - answer√√aqueous mixture, a substance is dissolved in water

homogeneous mixture - answer <a>A</a> mixture that is so evenly mixed that you can't see all of the different parts. Ex. Air, kool-aid

beterogeneous mixture - answer / / A mixture in which different materials can be seen easily.

Ex. soil, chocolate chip cookies

filtration - answer√√process to separate a solid "precipitate" from a liquid

sublimation - answer $\checkmark$  phase change from (s) --> (g)

empirical formula - answer√√A formula showing the lowest whole number ratio of atoms in a compound. Can't be reduced.

solid - answer√√A state of matter that has a definite shape and a definite volume

liquid - answer√√A form of matter that flows, has constant volume, and takes the shape of its container

Bromine (Br) and Mercury (Hg) - answer√√only two liquids on the periodic table

gas - answer - A state of matter with no definite shape or volume

gas - answer√√will uniformly and completely fill a container

# **Wayne State Chemistry Placement Exam Study Guide**

**Anthony S. Fauci** 

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