



Solubility Study Guide

Name _____

Teacher: _____

Date: _____

Period _____

References

Chemistry in the Community pp 45-65
Water properties: concept map
Solubility concept map
Solubility, graphing and analysis
Solubility Practice: A Curvy Subject
Study Solvent Lab

Vocabulary

Understand and apply the definitions or symbols of the following terms.

Colligative properties

Distillation

Filtration

Homogeneous

Mixtures

Nonpolar

Polar

Pure substances

Solubility

Solute

Solvent

Solution

Supersaturated

Saturated

Unsaturated

5+ and 5-

Targeted Skills

1. Describe the structure of water and relate its unusual physical properties to its structure, high boiling point, surface tension, high specific heat.
2. Draw the structure of water and relate its structure to the ability of water to act as a universal solvent.
3. Compare/contrast the solubilities of various ionic and covalent solutes.
4. Apply: Like dissolves like:
Polar solvents dissolve Polar and Ionic solutes
Nonpolar solvents dissolve Nonpolar solutes
5. Describe factors that affect the rate of dissolving, such as particle size, stirring, temperature, nature of the solid solutes.
6. Interpret information from a solubility curve (graph).

Make sure you remember from past tests?

Atomic structure

Significant figures

Accuracy & precision

Equations:(extension)

$$\text{Solution concentration} = \frac{\text{amount of solute}}{\text{amount solute} + \text{solvent}}$$

Test Breakdown

30 Multiple-Choice Problems (max.) – 60%

1 Free Response Problem chosen from the Free Response Test Bank – 10%

1 Open-Ended Problem chosen from the Open-Ended Test Bank– 10%

1 Performance Assessment – 20%

Study Guide Solubility And Concentration

Richard Whitten



Study Guide Solubility And Concentration:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

Chemical Principles Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text

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knowledge of basic pathophysiology as well as awareness of the appropriate diagnostic tests and treatments that are used to optimize care in the critically ill is essential. Since our first edition 7 years ago, new information crucial to the care and understanding of the critically ill patient has rapidly accumulated. Because this knowledge base crosses many different disciplines, a comprehensive multidisciplinary approach presenting the information is essential, similar to the multidisciplinary approach that is used to care for the critically ill patient. We have strived to provide this content in an easily digestible format that uses a variety of teaching tools to facilitate understanding of the presented concepts and to enhance information retention. To meet the demand to provide comprehensive and diverse education in order to understand the pathogenesis and optimum care of a variety of critical illnesses, we have substantially revised the prior topics in the first edition with updated information. We have also markedly expanded the number of topics covered to include acute lung injury and the acute respiratory distress syndrome, an expanded discussion of the physiology and operation of mechanical ventilation, obstetrical care in the ICU, neurosurgical emergencies, acute coronary syndromes, cardiac arrhythmias, role of whole body rehabilitation in the ICU, ethical conduct of human research in the ICU, and nursing care of the ICU patient. **NP-Midwife Specialty**

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