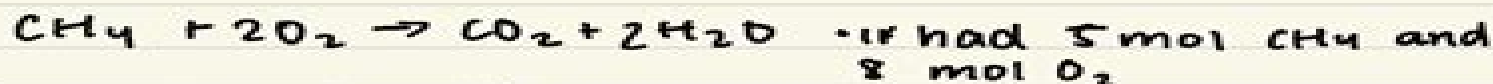


LIMITING REACTANT (thicked or grilled cheese)

- substance that is completely consumed in the chemical reaction, determines the amount of product that can be formed by reaction
 - ↳ identified by: # of moles of reactants, stoichiometry of balanced chemical equation
- when reactant runs out → reaction cannot proceed, dictates amount of product



$$5 \text{ mol CH}_4 \times \frac{2 \text{ mol O}_2}{1 \text{ mol CH}_4} = 10 \text{ mol O}_2$$

- you need 10 mol of O_2 to react with 5 mol of CH_4 but you only have 8 mol O_2 , so O_2 is the limiting reactant

$$8 \text{ mol O}_2 \times \frac{1 \text{ mol CH}_4}{2 \text{ mol O}_2} = 4 \text{ mol CH}_4 \quad \text{• you need 4 mol, have 5} \rightarrow \text{in excess}$$

LOOK AT MAX PRODUCTS

$$5 \text{ mol CH}_4 \times \frac{1 \text{ mol CO}_2}{1 \text{ mol CH}_4} = 5 \text{ mol CO}_2$$

$$8 \text{ mol O}_2 \times \frac{1 \text{ mol CO}_2}{2 \text{ mol O}_2} = 4 \text{ mol CO}_2$$

also known as theoretical yield

produces least amount of product so O_2 is the limiting reactant

THEORETICAL YIELD - calculated amount of product formed based on the amount of limiting reactant

ACTUAL YIELD - measured amount of product formed (what you get in the lab)

$$\text{PERCENT YIELD} = \frac{\text{ACTUAL YIELD}}{\text{THEORETICAL YIELD}} \times 100\%$$

Study Guide Stoichiometry Limiting Reactant Percent Yield

Iris Stovall



Study Guide Stoichiometry Limiting Reactant Percent Yield:

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The Complete Chemistry Study Guide and Note Cards and MCAT Konstantinos Papadopoulos,2012-07-06 *Study Guide to Accompany Basics for Chemistry* Martha Mackin,2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs

Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter,1993 Chemistry, Study Guide James E. Brady,John R. Holum,1993-02-11 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples *Barron's Chemistry Practice Plus: 400+ Online Questions and Quick Study Review* Barron's Educational Series,Mark Kernion,Joseph A. Mascetta,2022-07-05 Need quick review and practice to help you excel in Chemistry Barron's Chemistry Practice Plus features more than 400 online practice questions and a concise review guide that covers the basics of Chemistry Inside you ll find Concise review on the basics of Chemistry an excellent resource for students who want a quick review of the most important topics Access to 400 online questions arranged by topic for customized practice Online practice includes answer explanations with expert advice for all questions plus scoring to track your progress This essential guide is the perfect practice supplement for students and teachers Introductory Chemistry Study Guide Holder,2003-05

Basic Concepts of Chemistry, Study Guide Leo J. Malone,1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples

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