

Physical VS Chemical Changes

Directions: Identify which examples are physical or chemical changes. If it's physical, use a 'P'. If it's chemical, use a 'C'.

- P1. A pencil breaking in half.
- 6. Mixing baking soda and vinegar to cause the bubbling and fizzing.
- P4. Folding clothes after they come out of the dryer.
- C.5. When wood burns and you smell smoke.
- P.6. Clipping your fingernails.
- P7. Freezing water.
- C8. When gasoline in an engine combusts (burns) to create exhaust.
- P9. Changing the shape of a piece of Play-Doh.
- £ 10. Acid cooking a raw egg.
- £11. The leaves of a tree changes from green to brown in the Fall.
- P12. Smashing a bug.
- £13. When milk clumps up from spoiling.
- P14. A balloon popping.
- P15. Mixing sugar and water.
- £16. When the food that you eat digests.
- 17. Combining Mentos and Diet Coke to separate the CO₂.
- P18. A paper towel absorbing water.
- £19. Baking cookies in an oven.
- P20. Wadding up a piece of paper.

REMEMBER...

A <u>physical change</u> is a usually reversible change in the physical properties of matter, such as size or shape, but does not change the matter itself.

A <u>chemical change</u> is an irreversible change in one or more substances into entirely new substances with different properties. The <u>change</u> occurring from one substance to another is the <u>chemical reaction</u>.

Study Island Answer Key Physical And Chemical Change

Padhraic Smyth

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