1. The 3 particles of the atom are:

	Location in Atom?	Size of LOCATION?	Charge?	Mass?
Protons	nucleus	.0001%	+	1 anu
Neutrons	nucleus	.0001 %	neutral	lamer
Electrons	cloud	99.999 %		.00054ame

- 2. The number of protons in one atom of an element determines the atom's Auror 11.

 and the number of electrons determines the charge of an element.
- 3. The atomic number tells you the number of PYD TIMS in one atom of an element. It also tells you the number of PARTYTYS in a neutral atom of that element.
- 5. The <u>ONA ANOWE</u> of an element is the average mass of an element's naturally occurring atom, or isotopes, taking into account the <u>HOE NEXATORS</u> of each isotope.
- 6. The WARN of an element is the total number of protons and neutrons in the WARLS of the atom.

Study Packet Atomic Structure Answer Key

Peter L. Kresan, Reed Mencke, Frank Press, Raymond Siever

Study Packet Atomic Structure Answer Key:

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