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Subatomic Particles for Atoms, Ions and Isotopes-Practice

Use your periodic table to complete this worksheet.

	A tom/I on/I sotope name	Standard Atomic Notation	A tomic M ass	A tomic Number	Number of Protons	Number of Electrons	C harge	Number of Neutrons
1	Strontium (isotope)	90Sr	90	38	38	38	0	52
2	Beryllium (ion)	9Be ²⁺	9	4	4	2	+2	5
3	Selenium (ion)	⁷⁹ ₃₄ Se ²⁻	79	34	34	36	-2	45
4	A luminum (ion)	²⁷ ₁₃ Al ³⁺	27	13	13	10	+3	14
5	Helium (isotope)	³He	3	2	2	2	0	1
6	A rsenic (atom)	75 33 A S	75	33	33	33	0	42
7	Lithium (isotope)	¹ ₃ Li	11	3	3	3	0	8
8	Sodium (ion)	23Na1+	23	11	11	10	+1	12
9	M ercury (atom)	²⁰¹ ₈₀ Hg	201	80	80	80	0	121
10	Iron (isotope)	⁵³ ₂₆ Fe	53	26	26	26	0	27
11	Nitrogen (ion)	14N3-	14	7	7	10	-3	7
12	Nickel (ion)	$^{59}_{28}Ni^{3+}$	59	28	28	25	+3	31
13	C hlorine (isotope)	36Cl	36	17	17	17	0	19
14	X enon (atom)	¹³¹ ₅₄ Xe	131	54	54	54	0	77
15	Zinc (isotope)	$^{59}_{30}Zn$	59	30	30	30	0	29

*NOTE: when you are finished, you should have 3 atoms, 6 ions and 6 isotopes.

Subatomic Particle Gizmo Answer Sheet

Aidan Smith

Subatomic Particle Gizmo Answer Sheet:

Embracing the Track of Phrase: An Emotional Symphony within Subatomic Particle Gizmo Answer Sheet

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