

## CHAPTER 2

## STUDY GUIDE

## REPRESENTING MOTION

## Vocabulary Review

Write the term that correctly completes the statement. Use each term once.

average speed	instantaneous	origin	resultant
average velocity	position	particle model	scalar
coordinate system	instantaneous velocity	position	time interval
displacement	magnitude	position-time graph	vector
distance	motion diagram		

1. **Instantaneous velocity** \_\_\_\_\_ The speed and direction of an object at a particular instant is the \_\_\_\_\_.
2. **magnitude** \_\_\_\_\_ Another term given for the size of a quantity is the \_\_\_\_\_.
3. **position** \_\_\_\_\_ The \_\_\_\_\_ is the location of an object relative to an origin.
4. **time interval** \_\_\_\_\_ The formula  $t_2 - t_1$  represents \_\_\_\_\_.
5. **vector** \_\_\_\_\_ A \_\_\_\_\_ is a quantity with both magnitude and direction.
6. **average velocity** \_\_\_\_\_ Ratio of the change in position to the time interval during which the change occurred is the \_\_\_\_\_.
7. **coordinate system** \_\_\_\_\_ A system that defines the zero point of the variable you are studying is the \_\_\_\_\_.
8. **origin** \_\_\_\_\_ The zero point is also called the \_\_\_\_\_.
9. **position time graph** \_\_\_\_\_ A graph with time data on the horizontal axis and position data on the vertical axis is a \_\_\_\_\_.
10. **motion diagram** \_\_\_\_\_ A \_\_\_\_\_ shows a series of images showing the position of a moving object over equal time intervals.
11. **resultant** \_\_\_\_\_ A vector that represents the sum of two or more vectors is a \_\_\_\_\_.
12. **Particle model** \_\_\_\_\_ A simplified motion diagram that shows the object in motion as a series of points is a \_\_\_\_\_.

# Study Guide Vocabulary Review Write The Term

**S Nieto**



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oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni}^{2+} \rightarrow 3 \text{ Ag} + \text{Ni}$  oxidation reduction  $25 \text{ Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction  $26 \text{ O}_2$

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