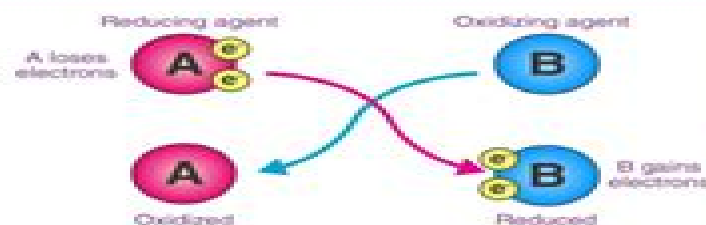


Redox reaction

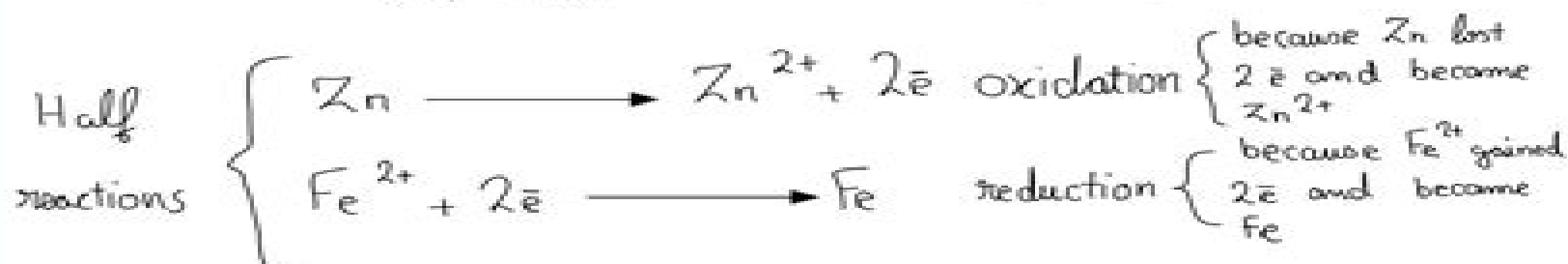
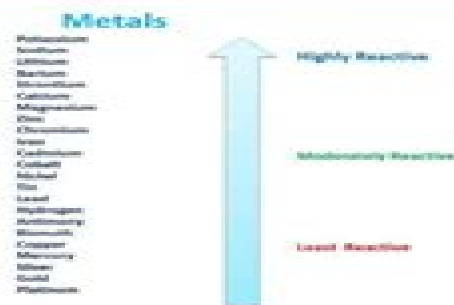
- A redox reaction is a chemical reaction where **both** oxidation and reduction take place
- An ion with positive charge is called Cations
- An ion with negative charge is called Anions
- Oxidation is a loss of electrons
- Reduction is a gain of electrons
- Not all the oxidation reactions involve oxygen

* The ion or molecule that accepts electrons is called the oxidizing agent - by accepting electrons it oxidizes other species.

* The ion or molecule that donates electrons is called the reducing agent - by giving electrons it reduced the other species.



- If a metal is more reactive than the metal corresponding to the ion there is going to be a reaction
- If a metal is less reactive than the metal corresponding to the ion there is **NOT** going to be a reaction



Study Guide Redox Reaction

Martha Mackin



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03. The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and

Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

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