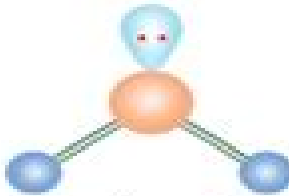
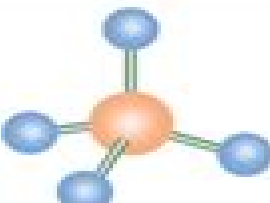
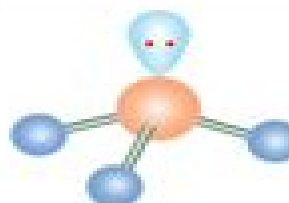
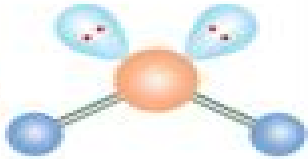
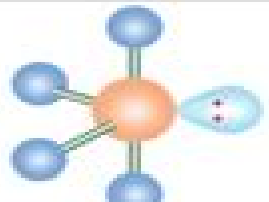
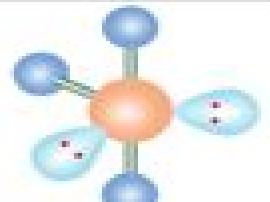
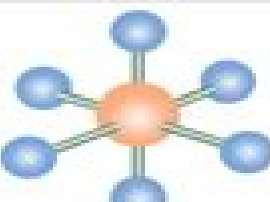
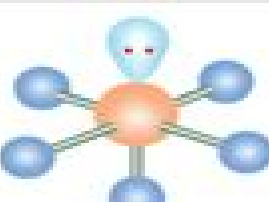
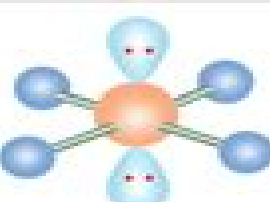
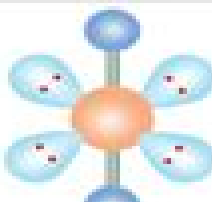


VSEPR Theory Chart

Number of Electron Groups	Lone Pairs = 0	Lone Pairs = 1	Lone Pairs = 2	Lone Pairs = 3	Lone Pairs = 4
2	 Linear				
3	 Trigonal Planar	 Angular or Bent			
4	 Tetrahedral	 Trigonal Pyramidal	 Angular or Bent		
5	 Trigonal Bipyramidal	 Seesaw	 T-shaped	 Linear	
6	 Octahedral	 Square Pyramidal	 Square Planar	 T-shaped	 Linear

Vsepr Theory Yahoo Answers

K Morrison



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