

Physical Science 3rd Quarter Exam Reviewer

I. Origin of Elements

Nucleosynthesis- formation of elements through nuclear reactions

Lesson 1.1 The Light Elements

- **Bigbang Nucleosynthesis** - the process where light elements were formed
- **Hydrogen and Helium**- the first light elements to be formed
- **Rapid drop of temperature** from 10^{32} to 10^9
- **Quarks and Gluons**- condensed into nucleons which formed the light elements

Lesson 1.2 The Heavy Elements

- **Stellar Nucleosynthesis** - the formation of heavy elements by fusion of lighter nuclei in the interior of stars.
- **Nebulas** - cloud of Hydrogen and Helium
- **Protostar** - mass of H and He
- **Stars can only produced elements up to Iron**

Lesson 1.3 Formation of Elements heavier than Iron

- **Supernova nucleosynthesis** - the explosive death of star; the creation of new elements based on the intensity of the burst
- **Extreme temperatures and Abundant neutrons** - Two key characteristics of Supernova explosion
- **Neutron capture** - when the nuclei captures a electrically neutral neutron to form heavier elements than iron

Lesson 1.4 Chemical Elements and Isotopes in the Universe

- **Isotopes** -is a form of an element that has the same atomic number of the original element but with different atomic mass or mass number.
- **Stable Isotope** - are not radioactive
- **Unstable Isotope** - are radioactive; decays and emits positron and neutrino to form a new element
- **Positron** - same mass as the electron but its charge is positive
- **Neutrino** - massless and has no charge

II. The Atomic Theory

Lesson 2.1 The Atomic Theory: From the Ancient Greeks up to the Present

- **Empedocles' Theory** - proposed that there were four fundamental substances. These substances constitute all matter.
- **Democritus' Theory** - he claimed that everything is made up of atoms; he called the very tiny pieces of matter "atomos" which means indivisible; the atom is the smallest possible piece of matter.
- **Aristotles' Theory**
accepted the ideas of Empedocles and even added his own idea that the four substances could be transformed into other substances. he associated the four fundamental elements- earth, water, air and fire with four qualities: hot, moist, cold and dry.
- **Akhenmyns' Theory of Matter**
based on Aristotles' theory of two pairs of opposed qualities and the four elements produced by pairwise conjunction of the four qualities. ; when these elements interchange qualities, they produce other substance.

Lesson 2.2 The Structure of the Atom

- **John Dalton** - proposed an atomic theory about indivisible building blocks of matter called atoms; atoms are spherical in shape and are always in motion.
- **Four Main Concepts of Dalton's Atomic Theory**
 - A. matter is composed of tiny, indivisible particles called atoms.
 - B. all atoms of the same element are identical in mass and properties.
 - C. compounds are formed by a combination of two or more different kinds of atoms.
 - D. a chemical reaction is a rearrangement of atoms.
- **Scientific Laws About Chemical Reactions**
 - A. 1. **Law of Conservation of Mass**
- Antoine Lavoisier in 1789
- This law states that the total mass in a chemical reaction remains constant.
 - B. **Law of Definite Proportions**
- Joseph Louis Proust in 1799

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