Name _____ Date ___ / ___ / ___ ATOMIC NUMBERS, PROTONS, NEUTRONS AND ELECTRONS

The number of protons in each atom of an element is its atomic number. The atomic number is also the number of electrons since atoms themselves are electronically neutral. Because neutrons and protons are almost the same mass, the total number of protons and neutrons in an atom is the atomic mass. Therefore, by subtracting the atomic number from the atomic mass you can calculate the number of neutrons.

This is why the mass number and atomic number, but not the number of neutrons, is shown in

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