Types of Chemical Reactions POGIL

Do atoms rearrange in predictable patters during chemical reactions?

why?

Recognizing patterns allows us to predict future behavior. Weather experts use patterns to predict dangerous storms so people can get their families to safety. Political analysts use patterns to predict election outcomes. Similarly, chemists classify chemical equations according to their patterns to help predict products of unknown but similar chemical reactions.

Model 1 - Types of Reactions

Set A	Set !
$4\text{Fe}(s) + 3O_{j}(g) \rightarrow 2\text{Fe}_{2}O_{j}(s)$	Mg0
$N_2(g) + 3H_2(g) \rightarrow 2NH_1(g)$	81.i,
$2SO_{s}(g) + O_{s}(g) \rightarrow 2SO_{s}(g)$	2H,
$MgO(s)+H_2O(l) \rightarrow Mg(OH)_2(aq)$	2800
$P_2O_3(g) + 3H_2O(l) \rightarrow 2H_2PO_4(aq)$	-2Na
$SO_3(g) + H_2O(l) \rightarrow H_2SO_4(aq)$	(NII
Set C	Set
$2\text{FeCl}_s(aq) + 3\text{Zn}(s) \rightarrow 2\text{Fe}(s) + 3\text{ZnCl}_s(aq)$	AgN
$2Al(NO_s)_s(aq) + 3Ca(s) \rightarrow 3Ca(NO_s)_s(aq) + 2Al(s)$	2111
$Mg(s) + CuSO_s(aq) \rightarrow MgSO_s(aq) + Cu(s)$	Nag
$2Al(s) + 6HCl(aq) \rightarrow 2AlCl_1(aq) + 3H_2(g)$	
$Cl_{j}(g) + 2NaBr(aq) \rightarrow 2NaCl(aq) + Br_{j}(l)$	FeS(
$ZnBr_2(aq) + F_3(g) \rightarrow ZnF_2(aq) + Br_2(l)$	HC

Set B	
$MgCO_{j}(s) \rightarrow MgO(s) + CO_{j}(g)$	
$8\text{Li}_2S(s) \rightarrow 16\text{Li}(s) + S_n(s)$	
$2H_{\downarrow}O(l) \rightarrow 2H_{\downarrow}(g) + O_{\downarrow}(g)$	
$2KClO_{s}(s) \rightarrow 2KCl(s) + 3O_{s}(g)$	
$2Na_sO_s(s) \rightarrow 2Na_sO(s) + O_s(g)$	
$(\mathrm{NH}_y)_2\mathrm{CO}_y(s) \rightarrow 2\mathrm{NH}_y(g) + \mathrm{H}_2\mathrm{O}(l) + \mathrm{CO}_2(g)$	
Set D	
AgNO _s (aq) + NaCl(aq) → AgCl(s) + NaNO _s (aq	×
$2HNO_{i}(aq) + Mg(OH)_{i}(aq) \rightarrow Mg(NO_{j})_{i}(aq) + 2H_{i}O(l)$	
$Na_sCO_s(aq) + CaCI_s(aq) \rightarrow CaCO_s(s) + 2NaCI_s(aq)$	
$FeS(s) + 2HCl(aq) \rightarrow H_sS(g) + FeCl_s(aq)$	
$HCl(aq) + NaOH(aq) \rightarrow H_2O(l) + NaCl(aq)$	
FeBr.(aq) + K.PO.(aq) → FePO.(s) + 3KBr(aq)	

1. Match each description below to one of the reactions sets (A, B, C, or D) from Model 1.

______ Ionic compounds dissolved in water switch partners.
One compound breaks into elements or smaller compounds
Two or more elements or compounds combine to form one product.
Part of an ionic compound is removed and replaced by a new element.

- 2. Define the following terms as they are commonly used in the English language:
 - a. Synthesis:
 - b. Decomposition:
 - c. Replacement:
 - 3.The four sets of chemical reactions shown in Model 1 have the following general names. Discuss within your group which name belongs to which set of chemical reactions. Write the name in the appropriate place in Model 1.

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Table of Contents Three Types Of Chemical Reactions Pogil Answer Key

- 1. Understanding the eBook Three Types Of Chemical Reactions Pogil Answer Key
 - The Rise of Digital Reading Three Types Of Chemical Reactions Pogil Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Three Types Of Chemical Reactions Pogil Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Three Types Of Chemical Reactions Pogil Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Three Types Of Chemical Reactions Pogil Answer Key
 - Personalized Recommendations
 - Three Types Of Chemical Reactions Pogil Answer Key User Reviews and Ratings
 - Three Types Of Chemical Reactions Pogil Answer Key and Bestseller Lists

- 5. Accessing Three Types Of Chemical Reactions Pogil Answer Key Free and Paid eBooks
 - Three Types Of Chemical Reactions Pogil Answer Key Public Domain eBooks
 - Three Types Of Chemical Reactions Pogil Answer Key eBook Subscription Services
 - Three Types Of Chemical Reactions Pogil Answer Key Budget-Friendly Options
- 6. Navigating Three Types Of Chemical Reactions Pogil Answer Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Three Types Of Chemical Reactions Pogil Answer Key Compatibility with Devices
 - Three Types Of Chemical Reactions Pogil Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Three Types Of Chemical Reactions Pogil Answer Key
 - Highlighting and Note-Taking Three Types Of Chemical Reactions Pogil Answer Key
 - Interactive Elements Three Types Of Chemical Reactions Pogil Answer Key
- 8. Staying Engaged with Three Types Of Chemical Reactions Pogil Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Three Types Of Chemical Reactions Pogil Answer Key
- 9. Balancing eBooks and Physical Books Three Types Of Chemical Reactions Pogil Answer Key
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Three Types Of Chemical Reactions Pogil Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Three Types Of Chemical Reactions Pogil Answer Key
 - Setting Reading Goals Three Types Of Chemical Reactions Pogil Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Three Types Of Chemical Reactions Pogil Answer Key
 - Fact-Checking eBook Content of Three Types Of Chemical Reactions Pogil Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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