Chapter 21: Nuclear Chemistry

- 1. What is a nuclide?
- What subatomic particle identifies a nuclide?
- 3. In carbon-12, what does the 12 represent?
- 4. Define mass defect.
- 5. What does mass defect represent?
- 6. What is Einstein's famous equation?
- 7. What kinds of elements have the highest nuclear binding energies?
- 8. How does the stability of a nuclide relate to the nuclear binding energy?
- On the band of stability, what is the neutron-proton ratio for low atomic number atoms?
- 10. How does the neutron-proton ratio change as the atomic number increases?
- 11. Unstable nuclides undergo what type of change?
- True or False: As a radioactive atom undergoes radioactive decay, the atom gives off large amounts of energy and increases its stability.
- 13. Define nuclear fission.
- 14. Define nuclear fusion.
- During radioactive decay, the nucleus disintegrates into a (lighter or heavier) nucleus that is (less stable or more stable)
- 16. How does alpha decay affect the nucleus?
- 17. How does beta decay affect the nucleus?
- 18. How does gamma radiation affect the nucleus?
- 19. How does positron emission affect the nucleus?
- 20. Which form of radiation has the greatest penetration power?
- Put the following in order from greatest mass to least mass: beta, alpha, gamma.
- 22. Alpha particles are (a) electrons (b)electromagnetic waves (c) helium nuclei
- 23. Beta particles are (a) electrons (b) electromagnetic waves (c) helium nuclei
- 24. What is the symbol for an alpha particle?
- 25. What is the symbol for a beta particle?
- 26. What is the symbol for a positron?
- 27. What is the symbol for a gamma ray?
- 28. Define half-life.
- 29. What can we say about a nuclide with a short half-life compared to one with a long half-life?
- 30. Define decay series.
- 31. What is the difference between natural transmutation and artificial transmutation?

Solve the following problems:

 How much of a 35.5 g sample of radon-222 is leftover after 30.4 days? The half-life of radon022 is 3.8 days.

Topic 12 Nuclear Chemistry Review Questions Answers

Michael Hoelscher

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