

Solutions & Dilutions Worksheet

Show all work on loose-leaf or notebook

- 1) If I add 25 mL of water to 125 mL of a 0.15 M NaOH solution, what will the molarity of the diluted solution be?
- 2) If I add water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity of the diluted solution be?
- 3) How much 0.05 M HCl solution can be made by diluting 250 mL of 10 M HCl?
- 4) I have 345 mL of a 1.5 M NaCl solution. If I boil the water until the volume of
- 5) How much water would I need to add to 500 mL of a 2.4 M KCl solution to make a 1.0 M solution?
- 6) Explain how you would make 450 mL of a 0.250 M NaOH solution.
- 7) To what volume will you have to dilute 30.0 mL of a 12 M HCl solution to make a 0.35 M HCl solution?
- 8) How many grams of calcium chloride will be needed to make 750 mL of a 0.100 M CaCl_2 solution?
- 9) Explain why this experimental procedure is incorrect: To make 1.00 L of a 1.00 M NaCl solution, I will dissolve 58.5 grams of sodium chloride in 1.00 L of water.
- 10) How many grams of beryllium chloride are needed to make 125 mL of a 0.050 M solution?
- 11) How many grams of beryllium chloride would you need to add to 125 mL of water to make a 0.050 molal solution?
- 12) The density of ethanol is 0.789 g/mL. How many grams of ethanol should be mixed with 225 mL of water to make a 4.5% (v/v) mixture?
- 13) Explain how to make at least one liter of a 1.25 molal ammonium hydroxide solution.
- 14) What is the molarity of a solution in which 0.45 grams of sodium nitrate are dissolved in 265 mL of solution.
- 15) What will the volume of a 0.50 M solution be if it contains 25 grams of calcium hydroxide?
- 16) How many grams of ammonia are present in 5.0 L of a 0.050 M solution?

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Ensheng Dong



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Unveiling the Power of Verbal Art: An Mental Sojourn through **Solutions Dilutions Worksheet Chemistry 2012 Tesccc**

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