(4-2)

Paraffin Family

stoichiometric chemical reaction

$$C_1H_{20} + 14 O_1 + 14(3.76) N_1 \rightarrow 9 CO_1 + 10 H_1O + 14(3.76) N_1$$

reaction with equivalence ratio $\phi = 0.7$

$$C_1H_{20} + (14/0.7) O_1 + (14/0.7)(3.76) N_1 \rightarrow 9 CO_1 + 10 H_2O + (14/0.7)(3.76) N_1 + 6 O_2$$

this reduces to

$$C_1H_{10} + 20 O_1 + 20(3.76) N_1 = 9 CO_1 + 10 H_1O + 20(3.76) N_1 + 6 O_1$$

Eqs. (2-55) and (4-1) $AF = m_m/m_s = N_mM_m/N_m/m_s = [(14)(4.76)(29)]/[(1)(128)] = 15.10$

(4-3)

(=)

Callin

isomer of octane

(b)

 C_pH_{20}

isomer of nonane

(c)

C.H.

isomer of octane

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