

# Test Solutions and Solubility SCH3U

Name: \_\_\_\_\_

15 K   14 T   9 C   10 A   Assessment of Learning (Product)

Multiple Choice. Circle the most correct answer. (15K, 5T)

1. How many of the following salts can be expected to be insoluble in water?

sodium sulfate   potassium chloride   ammonium nitrate   potassium chromate

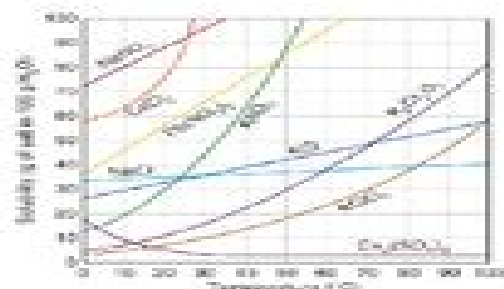
A) two   B) three   C) one   D) four   E) none

2. When solutions of silver nitrate and potassium chloride are mixed, which of the following will be present in the net ionic equation?

A)  $\text{KNO}_3(\text{aq})$    B)  $\text{K}^+$    C)  $\text{KCl}(\text{aq})$    D)  $\text{Cl}^-$    E) None of these

3. The net ionic equation for formation of an aqueous solution of  $\text{NiI}_2$  accompanied by evolution of  $\text{CO}_2$  gas via mixing solid  $\text{NiCO}_3$  and aqueous hydroiodic acid is \_\_\_\_\_.

- A)  $2\text{NiCO}_3(\text{s}) + \text{HI}(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g}) + 2\text{Ni}^{2+}(\text{aq})$   
 B)  $\text{NiCO}_3(\text{s}) + \text{I}^-(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g}) + \text{Ni}^{2+}(\text{aq}) + \text{HI}(\text{aq})$   
 C)  $\text{NiCO}_3(\text{s}) + 2\text{H}^+(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g}) + \text{Ni}^{2+}(\text{aq})$   
 D)  $\text{NiCO}_3(\text{s}) + 2\text{HI}(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g}) + \text{NiI}_2(\text{aq})$   
 E)  $\text{NiCO}_3(\text{s}) + 2\text{HI}(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l}) + \text{CO}_2(\text{g}) + \text{Ni}^{2+}(\text{aq}) + 2\text{I}^-(\text{aq})$



4. A beaker containing 80 grams of lead(II) nitrate,  $\text{Pb}(\text{NO}_3)_2$ , in 100 grams of water has a temperature of 30 °C. Approximately how many grams of the salt are undissolved, on the bottom of the beaker?

A) 14 grams   B) 66 grams   C) 80 grams   D) 20 grams

5. A solution in which no more solid can dissolve is called a \_\_\_\_\_ solution:

- A) solute   B) solution   C) electrolyte   D) immiscible   E) concentrated  
 F) soluble   G) solvent   H) unsaturated   I) dilute   J) non-electrolyte  
 K) saturated   L) miscible

# Solutions And Solubility Test Answers

**Daniel C. Tofan**



## **Solutions And Solubility Test Answers:**

*Excel Revise in a Month TEE Chemistry* Roy Skinner, 2003      **Class 9 Chemistry MCQ (Multiple Choice Questions)**

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## **Table of Contents Solutions And Solubility Test Answers**

1. Understanding the eBook Solutions And Solubility Test Answers
  - The Rise of Digital Reading Solutions And Solubility Test Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Solutions And Solubility Test Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Solutions And Solubility Test Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Solutions And Solubility Test Answers
  - Personalized Recommendations
  - Solutions And Solubility Test Answers User Reviews and Ratings
  - Solutions And Solubility Test Answers and Bestseller Lists
5. Accessing Solutions And Solubility Test Answers Free and Paid eBooks
  - Solutions And Solubility Test Answers Public Domain eBooks
  - Solutions And Solubility Test Answers eBook Subscription Services
  - Solutions And Solubility Test Answers Budget-Friendly Options
6. Navigating Solutions And Solubility Test Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Solutions And Solubility Test Answers Compatibility with Devices
  - Solutions And Solubility Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Solutions And Solubility Test Answers
  - Highlighting and Note-Taking Solutions And Solubility Test Answers
  - Interactive Elements Solutions And Solubility Test Answers
8. Staying Engaged with Solutions And Solubility Test Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Solutions And Solubility Test Answers
- 9. Balancing eBooks and Physical Books Solutions And Solubility Test Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Solutions And Solubility Test Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Solutions And Solubility Test Answers
  - Setting Reading Goals Solutions And Solubility Test Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solutions And Solubility Test Answers
  - Fact-Checking eBook Content of Solutions And Solubility Test Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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