

10.1

A student added 627 mg of hydrated sodium carbonate ($\text{Na}_2\text{CO}_3 \cdot x\text{H}_2\text{O}$) to 200 cm^3 of 0.250 mol dm^{-3} hydrochloric acid in a beaker and stirred the mixture.

After the reaction was complete, the resulting solution was transferred to a volumetric flask, made up to 250 cm^3 with deionised water and mixed thoroughly.

Several 25.0 cm^3 portions of the resulting solution were titrated with 0.150 mol dm^{-3} aqueous sodium hydroxide. The mean titre was 28.60 cm^3 of aqueous sodium hydroxide.

Calculate the value of x in $\text{Na}_2\text{CO}_3 \cdot x\text{H}_2\text{O}$.

Show your working.

Give your answer as an integer.

[7 marks]

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Table of Contents Titration Isa Gcse Section Paper

1. Understanding the eBook Titration Isa Gcse Section Paper
 - The Rise of Digital Reading Titration Isa Gcse Section Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Titration Isa Gcse Section Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Titration Isa Gcse Section Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Titration Isa Gcse Section Paper
 - Personalized Recommendations
 - Titration Isa Gcse Section Paper User Reviews and Ratings
 - Titration Isa Gcse Section Paper and Bestseller Lists

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 - Titration Isa Gcse Section Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Titration Isa Gcse Section Paper
 - Highlighting and Note-Taking Titration Isa Gcse Section Paper
 - Interactive Elements Titration Isa Gcse Section Paper
8. Staying Engaged with Titration Isa Gcse Section Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Titration Isa Gcse Section Paper
9. Balancing eBooks and Physical Books Titration Isa Gcse Section Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Titration Isa Gcse Section Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Titration Isa Gcse Section Paper
 - Setting Reading Goals Titration Isa Gcse Section Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Titration Isa Gcse Section Paper
 - Fact-Checking eBook Content of Titration Isa Gcse Section Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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