

Index Number:

Answer all the questions.

1. In a volumetric analysis, 24.80 cm^3 of $0.100 \text{ mol dm}^{-3}$ of a mineral acid, Y, neutralized 25.0 cm^3 of an alkaline solution containing 5.80 g of Na_2CO_3 per dm^3 .

(a) From the information provided, calculate the:

- (i) number of moles of Y;

- (ii) number of moles of Na_2CO_3 ;

ACS TEAM

- (iii) mole ratio of the acid to base in the analysis.

ZAMGIST

Waec Chemistry Practical Paper 3 Question And Answer 2014

Misra, Sanjay, Adewumi, Adewole



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This book delves into Waec Chemistry Practical Paper 3 Question And Answer 2014. Waec Chemistry Practical Paper 3 Question And Answer 2014 is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Waec Chemistry Practical Paper 3 Question And Answer 2014, encompassing both the fundamentals and more intricate discussions.

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- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Waec Chemistry Practical Paper 3 Question And Answer 2014.

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