Answer all the questions.

- In a volumetric analysis, 24.80 cm³ of 0.100 mol dm⁻³ of a mineral acid, Y, neutralized 25.0 cm³ of an alkaline solution containing 5.80 g of Na₂CO₃ per dm³.
 - (a) From the information provided, calculate the:
 - (i) number of moles of Y:

(ii) number of moles of Na₂CO₃;

(iii) mole ratio of the acid to base in the analysis.

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JR Anderson

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