

SIMPLE BINARY IONIC COMPOUNDS

PART A: Write the **formulas** for the following simple binary ionic compounds.

1. sodium sulfide Na_2S
2. magnesium nitride Mg_3N_2
3. potassium oxide K_2O
4. barium iodide BaI_2
5. potassium sulfide K_2S
6. silver sulfide Ag_2S
7. zinc nitride Zn_3N_2
8. barium fluoride BaF_2
9. lithium nitride Li_3N
10. rubidium oxide Rb_2O

11. calcium chloride CaCl_2
12. calcium phosphide Ca_3P_2
13. lithium fluoride LiF
14. magnesium bromide MgBr_2
15. aluminum oxide Al_2O_3
16. zinc oxide ZnO
17. aluminum iodide AlI_3
18. potassium phosphide K_3P
19. barium sulfide BaS
20. scandium phosphide ScP

PART B: Write the **names** for the following simple binary ionic compounds.

1. ZnI_2 Zinc iodide
2. MgBr_2 Magnesium bromide
3. Na_3N Sodium nitride
4. BaO Barium oxide
5. Na_2Si Sodium silicide
6. Zn_3N_2 Zinc nitride
7. Al_2O_3 Aluminum oxide
8. KBr Potassium bromide
9. Ru_3N_4 Ruthenium nitride
10. HfO_2 Hafnium oxide

11. CaO Calcium oxide
12. KF Potassium fluoride
13. Li_3P Lithium phosphide
14. SrS Strontium sulfide
15. ZnCl_2 Zinc chloride
16. K_2O Potassium oxide
17. KCl Potassium chloride
18. NaI Sodium iodide
19. La_2O_3 Lanthanum oxide
20. Cs_2S Cesium sulfide

Simple Binary Ionic Compounds Answer Key

Karin Nielsen-Saines



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