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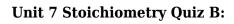
1.	2 KClO $_3 - 2$ KCl + 3 O $_2$ How many moles of oxygen are produced when 6.7 moles of KClO $_3$ decompose completely?		
jΔ,	10.1 mol	В	6.7 mol
C	1.0 mol	D	45 mol
2.	$2Na + 2H_2O \rightarrow 2NaOH + H_2$ Assuming H_2O is in excess, how Na?	many grams c	of hydrogen can be produced with 120 g of
A	2.6 g	В	690 g
C	45 g	D	5.2 g
3.	SiO ₂ + 3C → SiC + 2CO I need to add 8 moles of Carbon weigh?	to the reactio	n. How many grams of Carbon should I
,/%,	48 g	В	96 g
	0.67 g	D	2.67 g
4.	2Na + 2H ₂ O → 2NaOH+ H ₂ How many grams of hydrogen a	ire produced it	120 g of Na are available?
,Α,	2.6 g	В	690 g
C	45 g	D	5.2 g
5.	2CO + O ₂ > 2CO ₂		

How many liters of carbon dioxide are produced from 10L of carbon monoxide?

B 10

Unit 7 Stoichiometry Quiz B

Richard Bailey



The book delves into Unit 7 Stoichiometry Quiz B. Unit 7 Stoichiometry Quiz B is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Unit 7 Stoichiometry Quiz B, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Unit 7 Stoichiometry Quiz B
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 - o Chapter 3: Unit 7 Stoichiometry Quiz B in Everyday Life
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 - ∘ Chapter 5: Conclusion
- 2. In chapter 1, this book will provide an overview of Unit 7 Stoichiometry Quiz B. This chapter will explore what Unit 7 Stoichiometry Quiz B is, why Unit 7 Stoichiometry Quiz B is vital, and how to effectively learn about Unit 7 Stoichiometry Quiz B.
- 3. In chapter 2, this book will delve into the foundational concepts of Unit 7 Stoichiometry Quiz B. This chapter will elucidate the essential principles that need to be understood to grasp Unit 7 Stoichiometry Quiz B in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Unit 7 Stoichiometry Quiz B in daily life. The third chapter will showcase real-world examples of how Unit 7 Stoichiometry Quiz B can be effectively utilized in everyday scenarios.
- 5. In chapter 4, the author will scrutinize the relevance of Unit 7 Stoichiometry Quiz B in specific contexts. The fourth chapter will explore how Unit 7 Stoichiometry Quiz B is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, this book will draw a conclusion about Unit 7 Stoichiometry Quiz B. The final chapter will summarize the key points that have been discussed throughout the book.
 - This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Unit 7 Stoichiometry Quiz B.

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