

charge on the carbon makes it polar.

Electrophile is an **electron-pair donor**.
Electrophile is an electron pair to somewhere without enough electrons.
Electrophiles are **electron pair donors** that can react with halogenoalkanes.
 OH^- and NH_3 are all nucleophiles that can react with halogenoalkanes.



Halogenoalkanes Can Undergo Nucleophilic Substitution Reactions

Halogenoalkanes can react with a polar molecule by kicking out the functional group and taking its place.
This is called a **nucleophilic substitution reaction**. It works like this:



X^- stands for one of the halogens (F, Cl, Br or I).
 Nu^- stands for a nucleophile.

The lone pair of electrons on the nucleophile attacks the δ^+ carbon.

The C-X bond breaks.

The halogen leaves, taking both electrons with it. A new bond forms between the carbon and the nucleophile.

Product of these reactions depends on what the nucleophile is...

Alkanes React with Hydroxides to Form Alcohols

Alkanes, bromoethane can react to form ethanol. You have to use warm sodium or potassium hydroxide — it's a nucleophilic substitution reaction.

This is sometimes **hydrolysis**, because exactly the same will happen with (Hydrolysis is splitting a molecule).

Unit 3 Section 1 Quiz Chemistry

A Loxley



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