Acid-Base Chemistry

acids

- pH less than 7
- Turn litmus paper red
- Taste sour
- Feel irritating/corrosive
- Include fruit juices, soda, and coffee

bases

- pH greater than 7
- · Turn litmus paper blue
- Taste bitter or soapy
- · Feel slippery
- Include baking soda, ammonia, and soap



neutralization reaction example: HCl + NaOH → NaCl + H₂O

Unit 14 Acid And Bases

Gary M. Rand

Unit 14 Acid And Bases:

A Laboratory Textbook of Anatomy and Physiology: Cat Version Donnersberger, 2009-03-02 Thoroughly updated throughout and now incorporating a full color design and art program the ninth edition of A Laboratory Textbook of Anatomy and Physiology provides students with an accessible comprehensive introduction to A P It is specifically designed for the laboratory portion of a one or two term course in anatomy and physiology for students planning a health science allied health or health related career The texts 15 integrated units use the cat as the dissection animal while also emphasizing the human anatomy This classic text is a proven must have resource and learning tool for the A P lab Experimental Organic Chemistry Daniel R. Palleros, 2000-02-04 This cutting edge lab manual takes a multiscale approach presenting both micro semi micro and macroscale techniques. The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion A Laboratory Textbook of Anatomy and Physiology Anne B. Donnersberger, Anne Lesak Scott, 2005-10 At last a brand new fetal pig version of the classic laboratory textbook by Donnersberger and Lesak Scott This new book is the ideal lab text for a one or two term course in anatomy and physiology for students planning a health science or health related career Featuring fifteen integrated units each consisting of a Purpose Objectives Materials Procedures Self Test Case Studies and Short Answer Questions this comprehensive lab text makes an ideal companion to any current anatomy and physiology text or it can be used as both a main text and lab manual **Elements of Environmental Engineering** Kalliat T. Valsaraj, Elizabeth M. Melvin, 2009-06-09 Revised updated and rewritten where necessary but keeping the clear writing and organizational style that made previous editions so popular Elements of Environmental Engineering Thermodynamics and Kinetics Third Edition contains new problems and new examples that better illustrate theory The new edition contains examples with practical flavor such as global warming ozone layer depletion nanotechnology green chemistry and green engineering With detailed theoretical discussion and principles illuminated by numerical examples this book fills the gaps in coverage of the principles and applications of kinetics and thermodynamics in environmental engineering and science New topics covered include Green Chemistry and Engineering Biological Processes Life Cycle Analysis Global Climate Change The author discusses the applications of thermodynamics and kinetics and delineates the distribution of pollutants and the interrelationships between them His demonstration of the theoretical foundations of chemical property estimations gives students an in depth understanding of the limitations of thermodynamics and kinetics as applied to environmental fate and transport modeling and separation processes for waste treatment His treatment of the material underlines the multidisciplinary nature of environmental engineering This book is unusual in environmental engineering since it deals exclusively with the applications of chemical thermodynamics and kinetics in environmental processes The book s multimedia approach to fate and transport modeling and in pollution control design options provides a science and engineering

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