

## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- Row 1:** A (1,1), B (1,13)
- Row 2:** C (2,1), D (2,10), E (2,11)
- Row 3:** F (3,2), G (3,7), H (3,9), K (3,13)
- Row 4:** L (4,2), I (4,11)
- Row 5:** J (5,1), Z (5,3), M (5,7), N (5,9), O (5,11), P (5,13)
- Row 6:** Q (6,1), R (6,3), S (6,7), T (6,8), U (6,9)

Below the main grid, there is a separate row of cells with the following letters:

- Row 7:** V (7,1), W (7,7)
- Row 8:** X (8,1), Y (8,7)

Arrows point from the letter 'R' in the main grid (6,3) to the letters 'V' (7,1) and 'X' (8,1) in the separate row.

List elements in *alphabetical order* when more than 1 element answers a question.

- \_\_\_\_\_ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- \_\_\_\_\_ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- \_\_\_\_\_ 3) What is the letter of the element whose atomic number is closest to the element labeled "E"?
- \_\_\_\_\_ 4) Which letter labels the halogen element with the lowest atomic number?
- \_\_\_\_\_ 5) Which of these elements is the least reactive? **A)** "D" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- \_\_\_\_\_ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "W" **E)** "Y"

continued on other side

# Using The Periodic Table Answer Key

**Neil D. Jespersen, Pamela Kerrigan**



## **Using The Periodic Table Answer Key:**

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