

Chemical Bonding Review Worksheet

1. Which part of the atom is responsible for chemical bonding?

2. What are valence electrons (be specific)?

3. How many valence electrons do most atoms need to have a complete outer shell and be happy? _____
4. Which two elements only need two valence electrons to be happy?

5. Why do the elements you named in #4 only need two valence electrons?

6. How many valence electrons do elements in Group 1, the Alkali Metals, have?

7. How many valence electrons do elements in Group 2, the Alkaline Earth Metals, have?

8. What is an ion?

9. If an element gives away an electron, will it form a positive ion or a negative ion?

10. If an element gains an electron, will it form a positive ion or a negative ion?

11. What is the difference in a cation and an anion?

12. How do ionic bonds form?

13. How do covalent bonds form?

14. Bond the following atoms. Determine if they are ionic or covalent, circle your choice. Show the valence electrons and how they are either shared between the atoms or how they are transferred between atoms. Then write the chemical formula in the space provided.

Ionic or covalent		Ionic or covalent	
C	Cl	Mg	Cl
Formula _____		Formula _____	

Valence Electrons Chemistry If8766 Answers

R Sanford



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