Valence Electrons

How many electrons are present in the following elements? [1]

Carbon ___ B Chlorine ___ a Indium ___ a Calcium ___

Using the periodic table, determine the number of valance electrons for the 2 elements listed below.

Barium In Iodine Carbon In Phosphorous

Circle the valence electrons only. What is the atomic number?









Atomic number(Z) = Atomic number(Z) = Atomic number(Z) = Atomic number(Z) =

Fill in the table with correct values. 4

| Element | Ion symbol | Number of Protons | Number of electrons | Charge | Ion type |
|-----------|------------|----------------------|------------------------|--------|----------|
| Fluorine | F- | | | - 1 | Anion |
| | | 53 | 54 | | |
| | | 16 | | -2 | 1 |
| Potassium | | | | +1 | |
| Calcium | | | | ., | |

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Ensheng Dong

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