

Catalyst
by Whirlpool

SELECT CYCLE

WASH TYPE

- | | |
|--|--|
| ☒ Heavy Duty
WASH TYPE | ☐ Ultra
Hardwash
WASH TYPE |
| ☐ Regular
WASH TYPE | ☐ Express Wash
WASH TYPE |
| ☐ Personal
Wash
WASH TYPE | ☐ Whisper
Washes
WASH TYPE |
| ☐ Handwash
Cycle
WASH TYPE | ☐ Custom Set
WASH TYPE |

OPTIONS

- ☒ Catalyst
- ☐ Soak
- ☐ Drain / Spin
- ☐ Rinse / Spin
- ☒ 2nd Rinse
- ☐ Extra Options

MODIFIERS

☒ Auto
Timer
Control



WATER
LEVEL

☐ Hot Water
Wash Cold

☐ Cold Water
Wash Cold

WASH/WARM
TEMP

☒ Extended
Tub Washing

56

☐ Delay Time

☐ Adaptive Time

ADAPTATION
TIME

DRYER STATUS

☐ Idle

☐ Heating

☐ Stopped

☐ Cycle Complete

☐ Lock

☒ Quiet

☐ Off

END OF CYCLE
SIGNAL

START

STOP

Heavy Duty • 4000 Spin Capacity • Ultra Quiet Pad • 2 Speed Motor With 4 Speed Combination

Whirlpool Calypso Washer Error Codes

Douglas Emley

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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 net ionic equation $\text{H}^+\text{aq} + \text{HCO}_3^-\text{aq} \rightarrow \text{H}_2\text{O} + \text{CO}_2\text{g}$ here is a video on spectator ions
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web the six Na^+ ions and the six Cl^- ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+}\text{aq} + 2\text{PO}_4^{3-}\text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2\text{s}$

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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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 $\text{AgNO}_3\text{aq} + \text{KClaq} \rightarrow \text{AgCl s} + \text{KNO}_3\text{aq}$ total ionic $\text{Ag}^+\text{aq} + \text{NO}_3^-\text{aq} + \text{K}^+\text{aq} + \text{Cl}^-\text{aq} \rightarrow \text{AgCl s} + \text{K}^+\text{aq} + \text{NO}_3^-\text{aq}$ net ionic $\text{Ag}^+\text{aq} + \text{Cl}^-\text{aq} \rightarrow \text{AgCl s}$ 2

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+\text{aq} + \text{OH}^-\text{aq} \rightarrow \text{H}_2\text{O} + \text{l}$ the spectator ions in KOHaq and HNO_3aq are K^+aq and NO_3^-aq how to identify and list the

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anions an ion with a net charge of zero is electrically neutral

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web the ag aq and cl aq ions become agcl s but the na aq ions and the no 3 aq ions stay as na aq ions and no 3 aq ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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