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1. The atomic number of the element _____, the element specified under previous letter
2. The atomic number of the element _____, the element specified under previous letter
3. The number of π in $2\pi^2$ is referred to as _____, which represents number of atomic orbitals
4. The number of π in $2\pi^2$ is referred to as _____, and the number of orbitals _____

WILLIAM L. BRYAN

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10

Answer: (C) 1 mole = 6.02 $\times 10^{23}$ atoms and 1 gram of H = 6.02 $\times 10^{23}$ atoms

1. How many years do you work?

 1.406×10^{-2}

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